

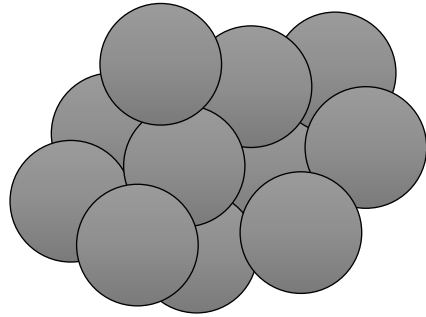
Understanding Phonon-Related Properties in Metal-Organic Frameworks for Controlling Their Mechanical and Thermal Characteristics

Tomas Kamencek

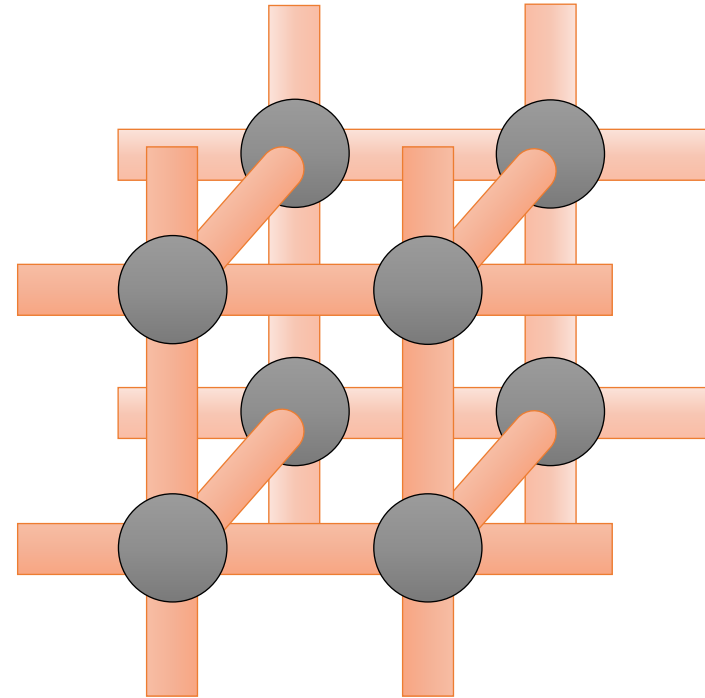
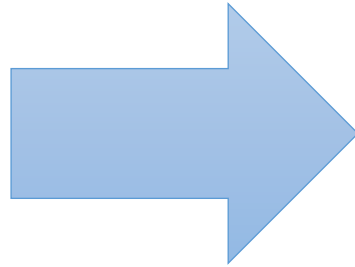
Institute of Solid State Physics &
Institute of Physical and Theoretical Chemistry

Metal-Organic Frameworks (MOFs)

Metal (oxide) nodes



Organic linkers



Science, 2013, 341, 1230444

Microporous Mesoporous Mater., 2004, 73,3

THOUSANDS of m² per gramme of the material

Gas storage

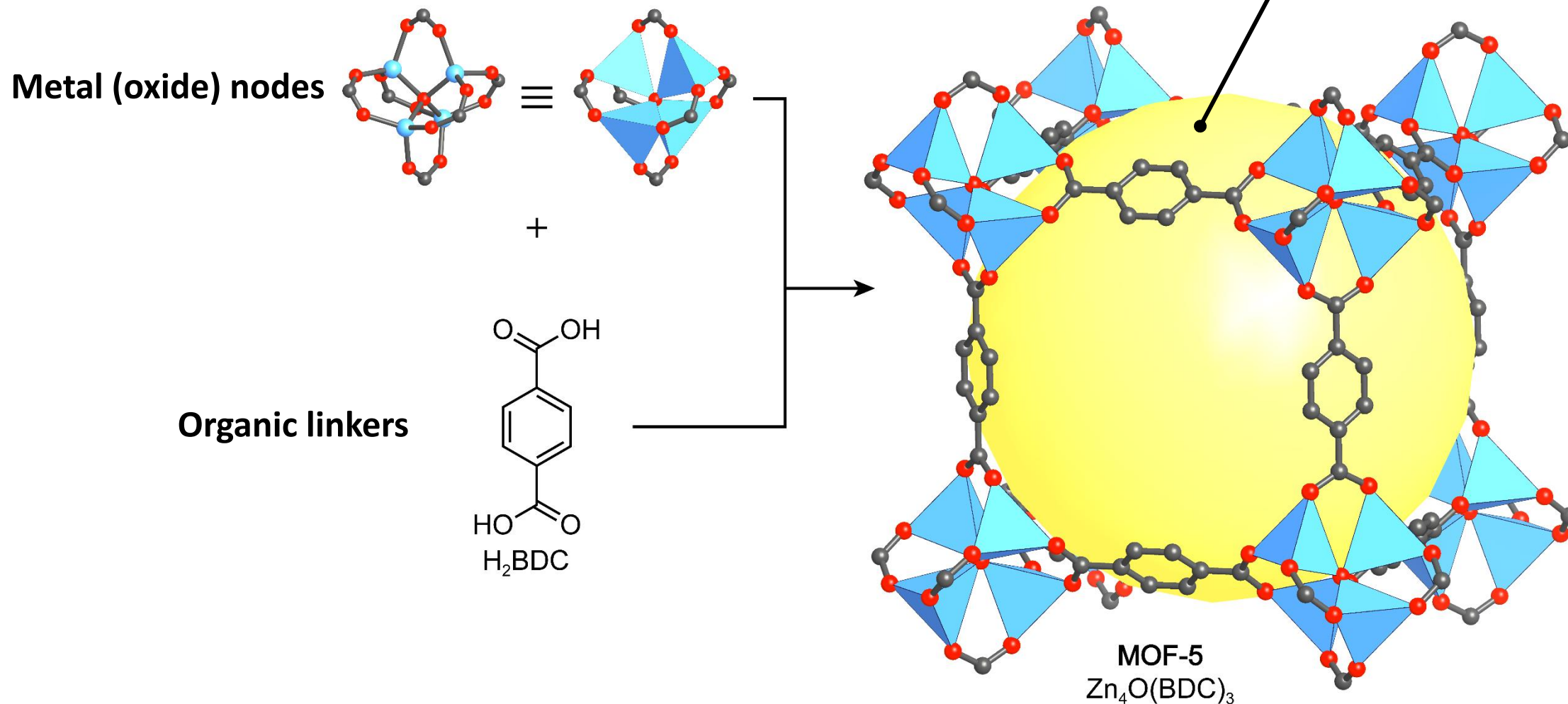
Catalysis

Gas separation

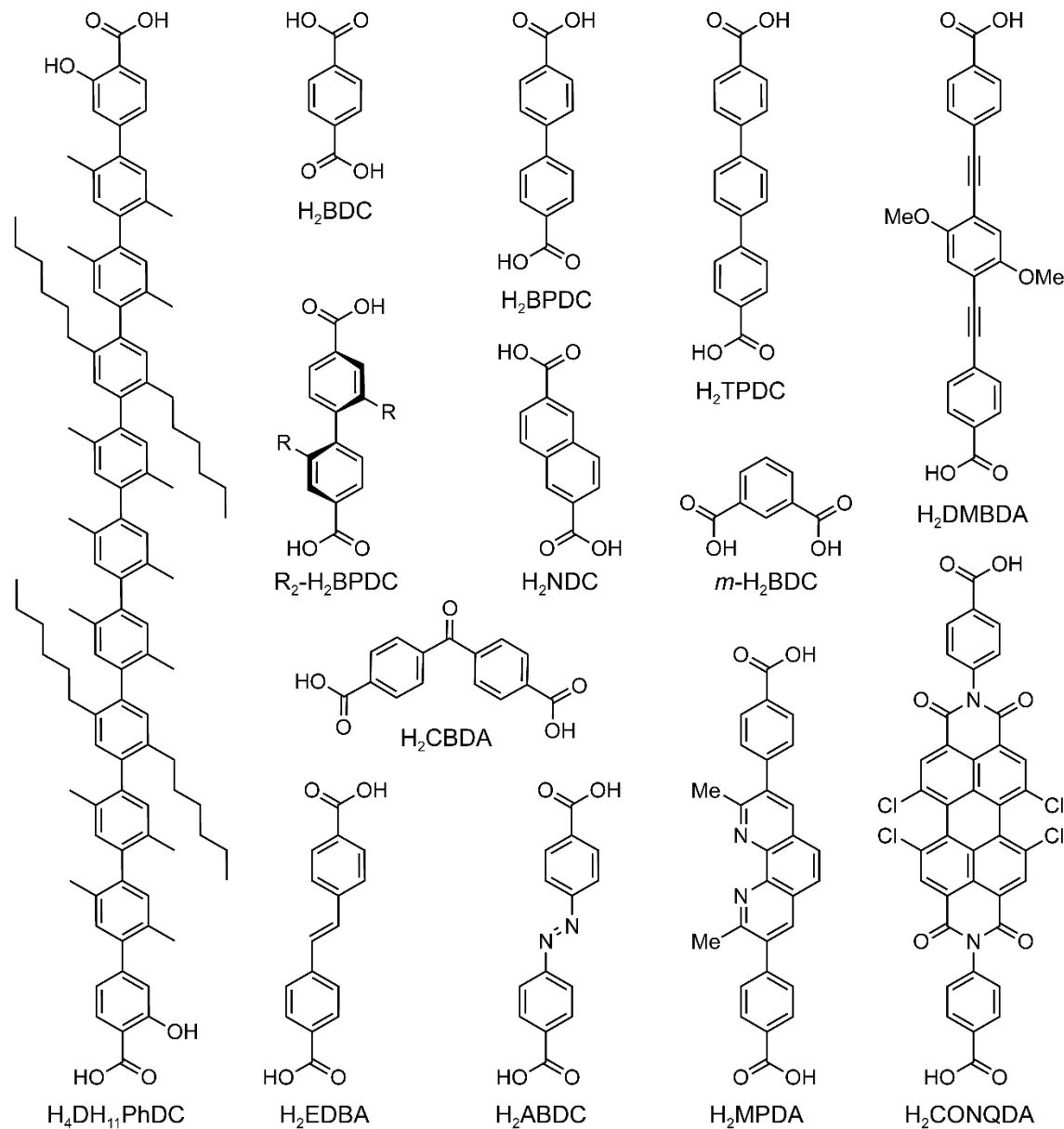
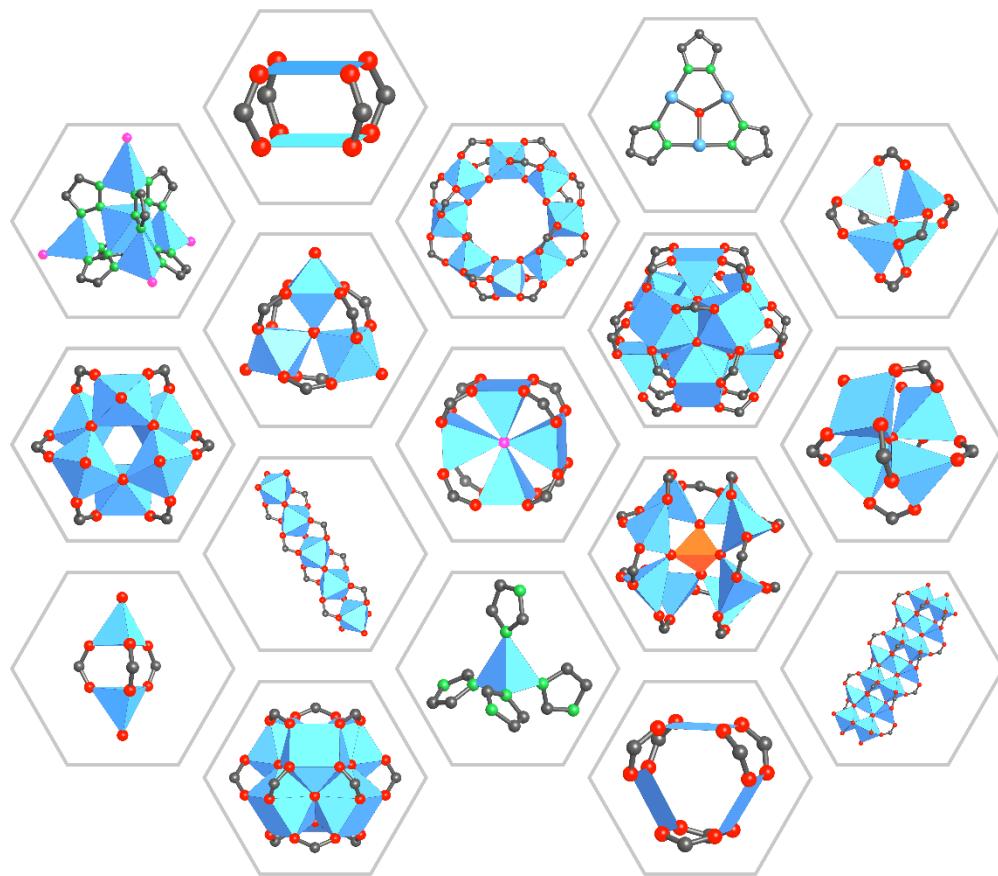
Encapsulation

Functional decives

A Prototypical MOF: MOF-5



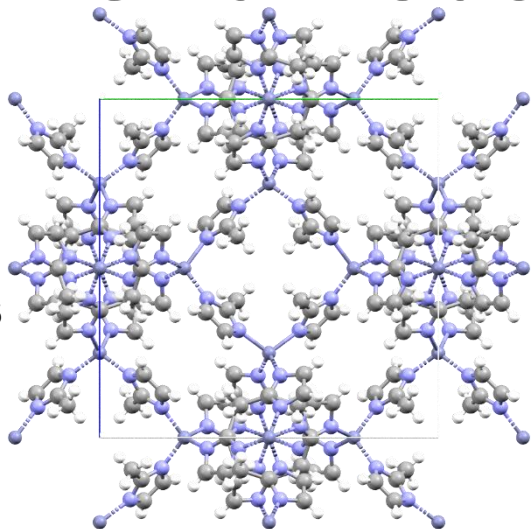
Nodes + Linkers



Some Famous MOFs

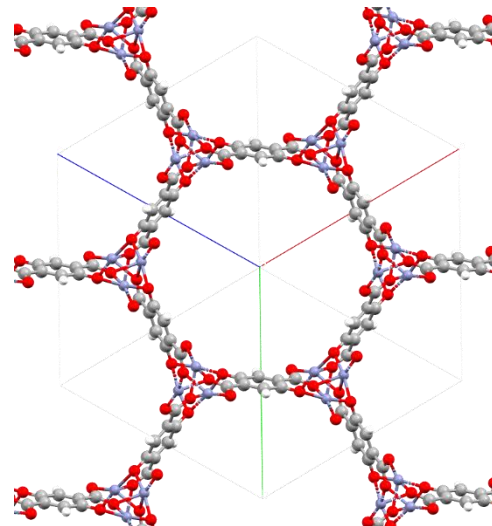
ZIF-8

Phys. Rev. Lett.,
2012, 108, 1-6



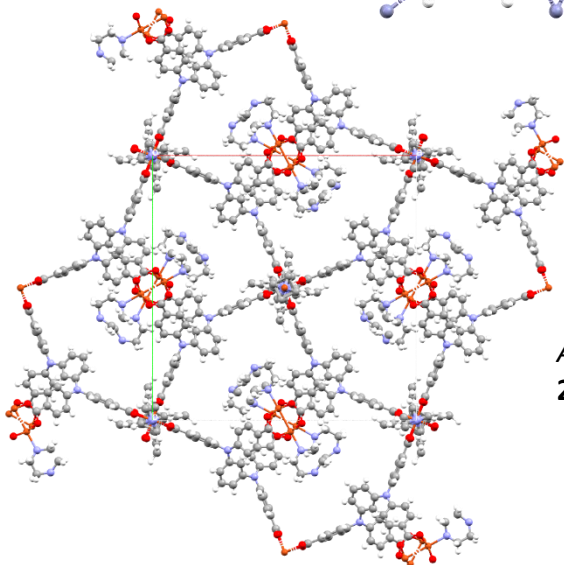
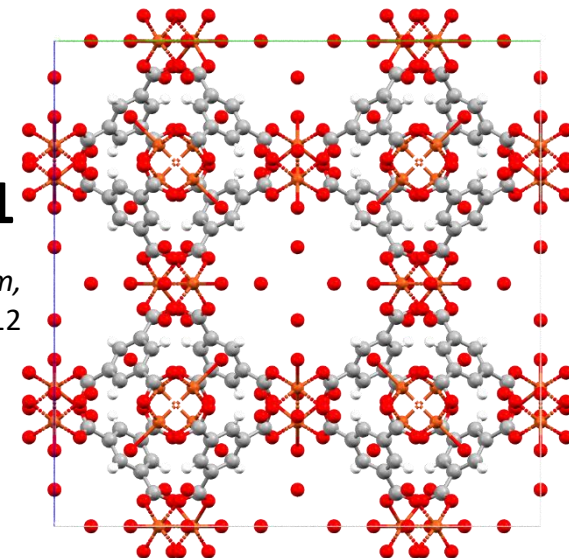
MOF-74

J. Mater. Chem. A,
2015, 3, 986-995



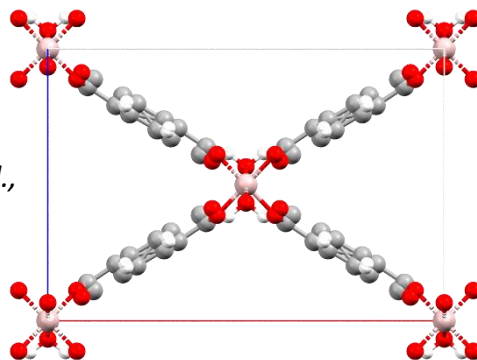
HKUST-1

CrystEngComm,
2016, 18, 4303-4312



DUT-90

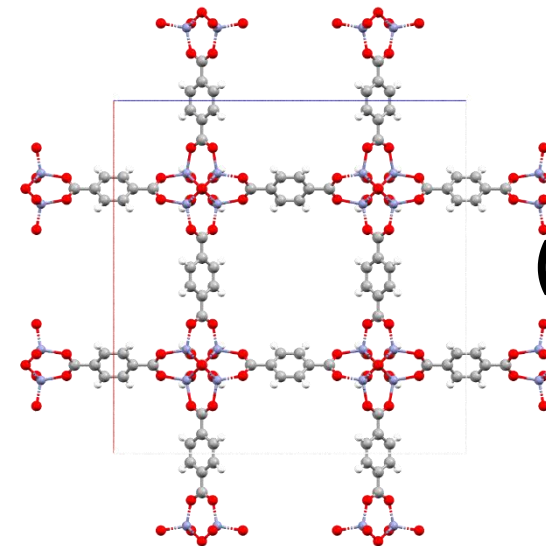
Angew. Chemie –Int. Ed.,
2018, 57, 13780-13783



MIL-53

Phys. Rev. Lett.,
2012, 109, 1-5

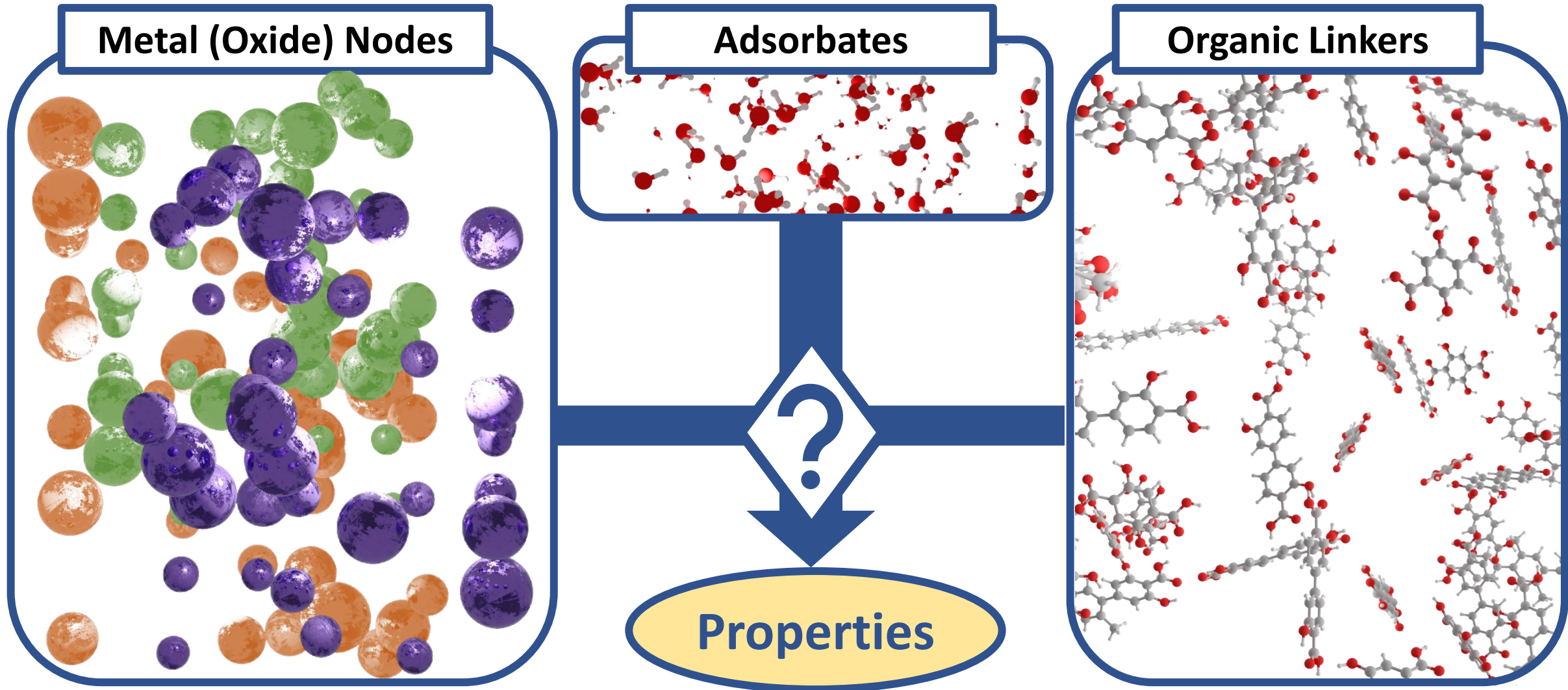
J. Chem. Phys.,
2013, 138, 174703



MOF-5 (=IRMOF-1)

J. Chem. Phys.,
2017, 146, 184705

Properties in Metal-Organic Frameworks



Relevant Properties of MOFs

Phonons determine many of the relevant properties of a MOF!

Γ -Phonons

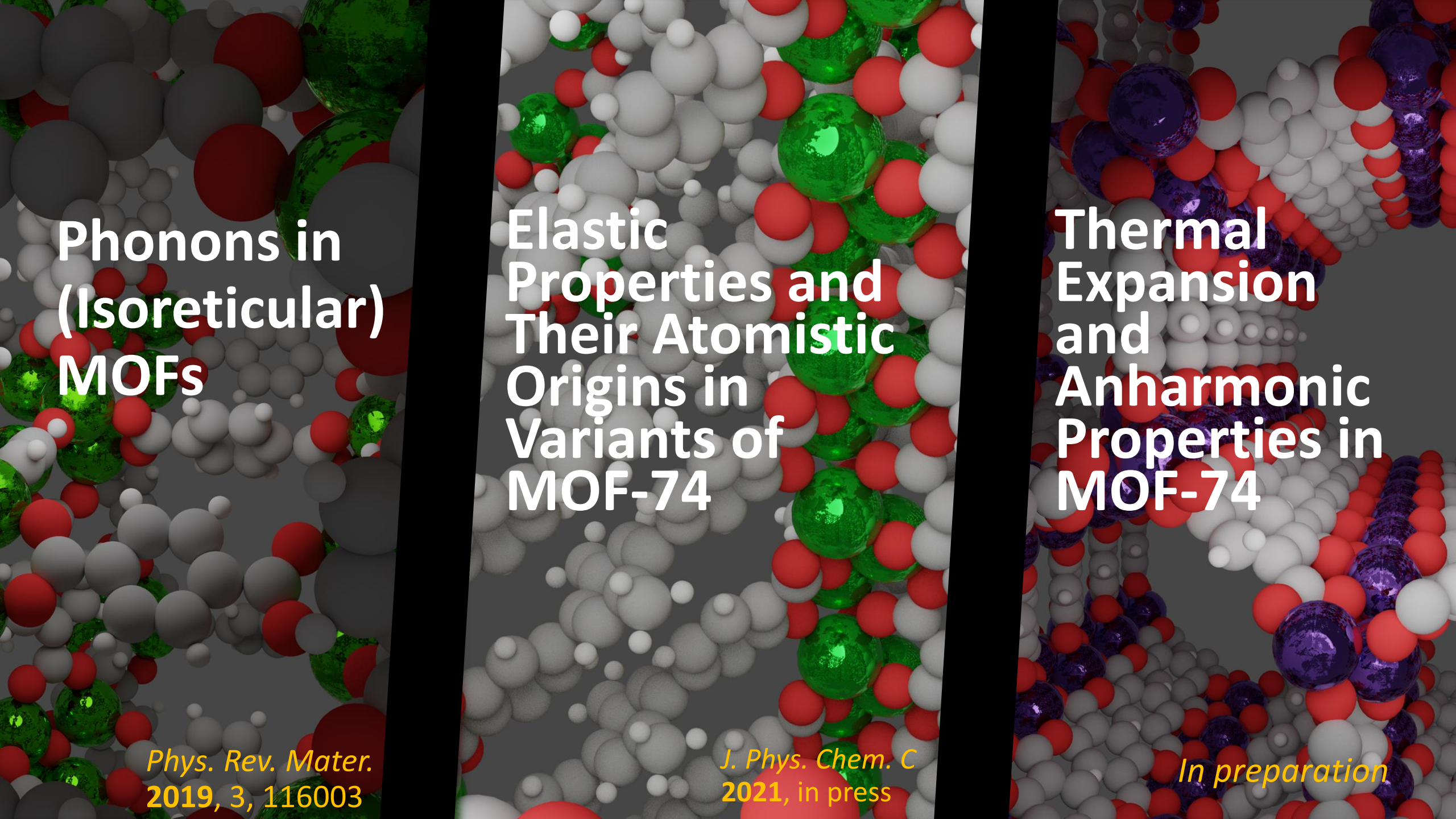
- **Characterisation:**
Raman/IR
Spectroscopy
- **Phase
Identification**

Band Structures

- **Vibrational Free
Energy / Entropy**
- **Elastic Constants**
- **Band Dispersion
(Group Velocities)**

Anharmonicity

- **Thermal
Expansion**
- **Phonon Lifetimes**
- **Thermal
Conductivity**



Phonons in
(Isoreticular)
MOFs

Phys. Rev. Mater.
2019, 3, 116003

Elastic
Properties and
Their Atomistic
Origins in
Variants of
MOF-74

J. Phys. Chem. C
2021, in press

Thermal
Expansion
and
Anharmonic
Properties in
MOF-74

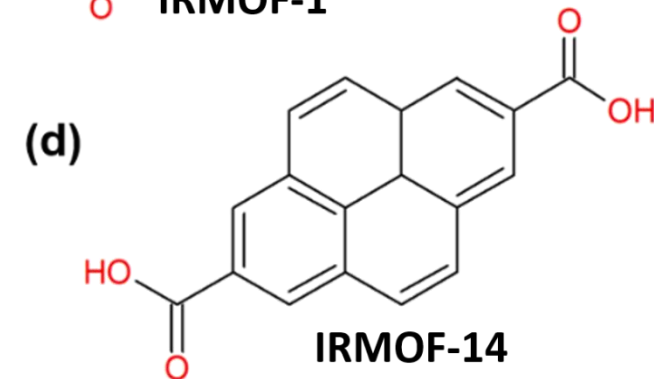
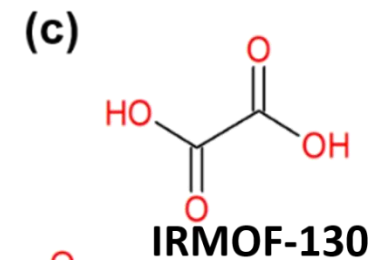
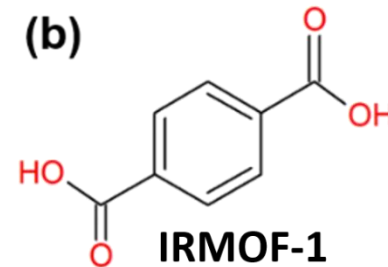
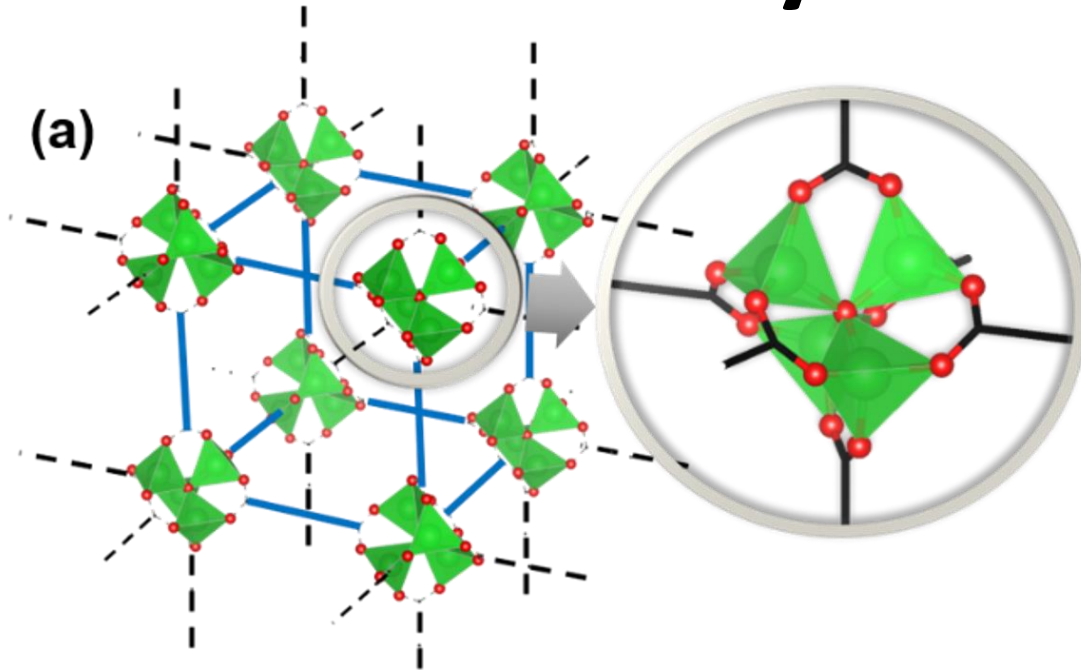
In preparation

The background of the slide is a 3D visualization of a Metal-Organic Framework (MOF) structure. It consists of a dense packing of spheres in three colors: grey, red, and green. The grey spheres are the most numerous and form a complex, interconnected network. The red and green spheres are interspersed within this network, representing different types of atoms or functional groups. The overall appearance is that of a porous, crystalline material.

Phonons in (Isoreticular) MOFs

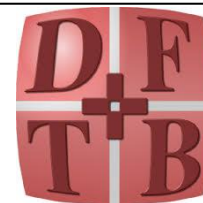
Kamencek, Bedoya-Martínez, and Zojer; *Phys. Rev. Mater.* 2019, 3, 116003

Goals of the Study



- **Systematic analysis** of the phonon band structures as a function of
 - The metal ions (Mg^{2+} , Ca^{2+})
 - The linker (\rightarrow IRMOF-1, IRMOF-130, IRMOF-14)
- In which **frequency regimes** can one find certain phonon modes?

Methods



PHONOPY

<https://atztogo.github.io/phonopy/>

Kresse and Hafner, *Phys. Rev. B*, **1993**, 47 (1), 558
 Kresse and Hafner, *Phys. Rev. B*, **1994**, 49 (20), 14251

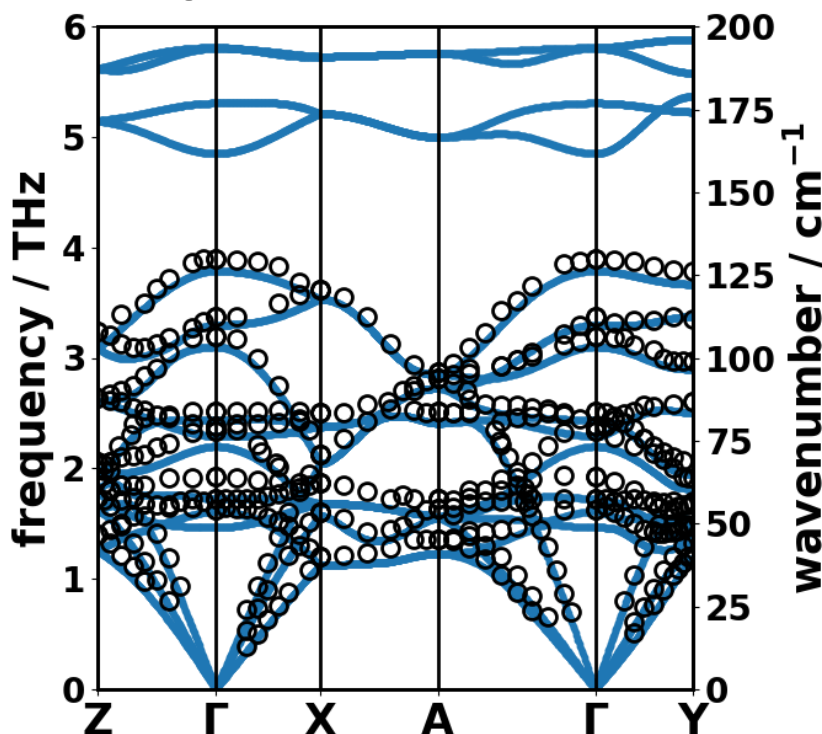
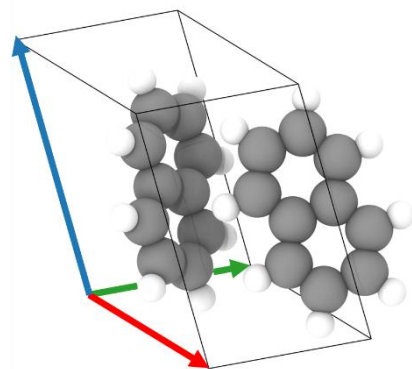
Aradi et al. *J. Phys. Chem. A*, **2007**, 111 (26), 5678

Togo and Tanaka, *Scr. Mater.*, **2015**, 108, 1

D3-BJ correction:

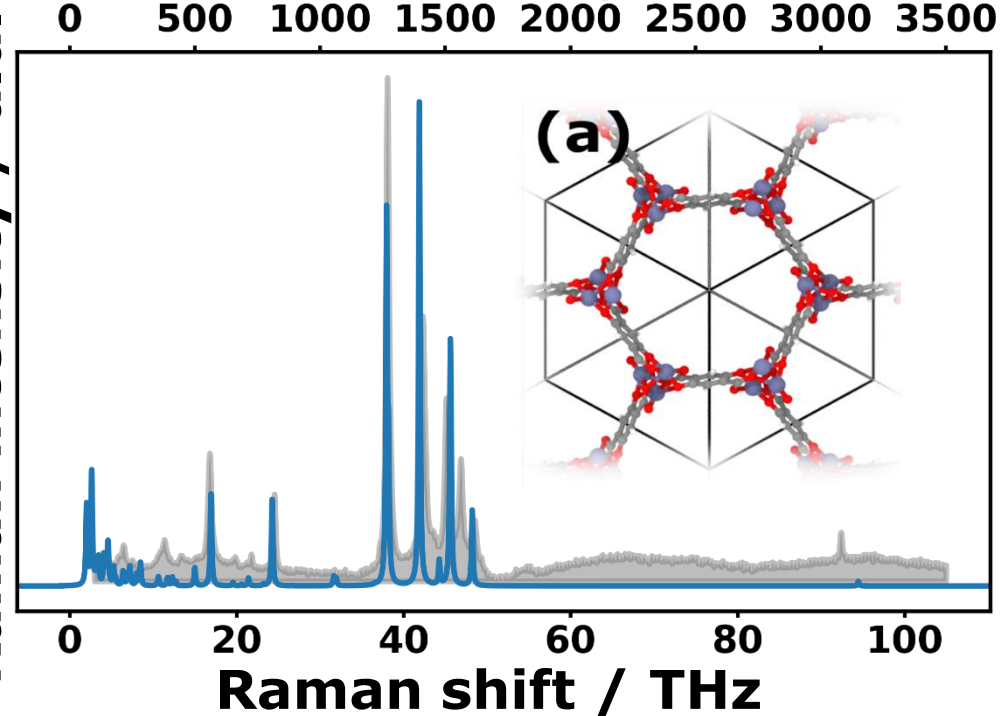
Grimme et al., *J. Chem. Phys.*, **2010**, 132 (15), 154104; Grimme et al., *J. Comput. Chem.*, **2011**, 32 (7), 1456

Crystalline Naphthalene



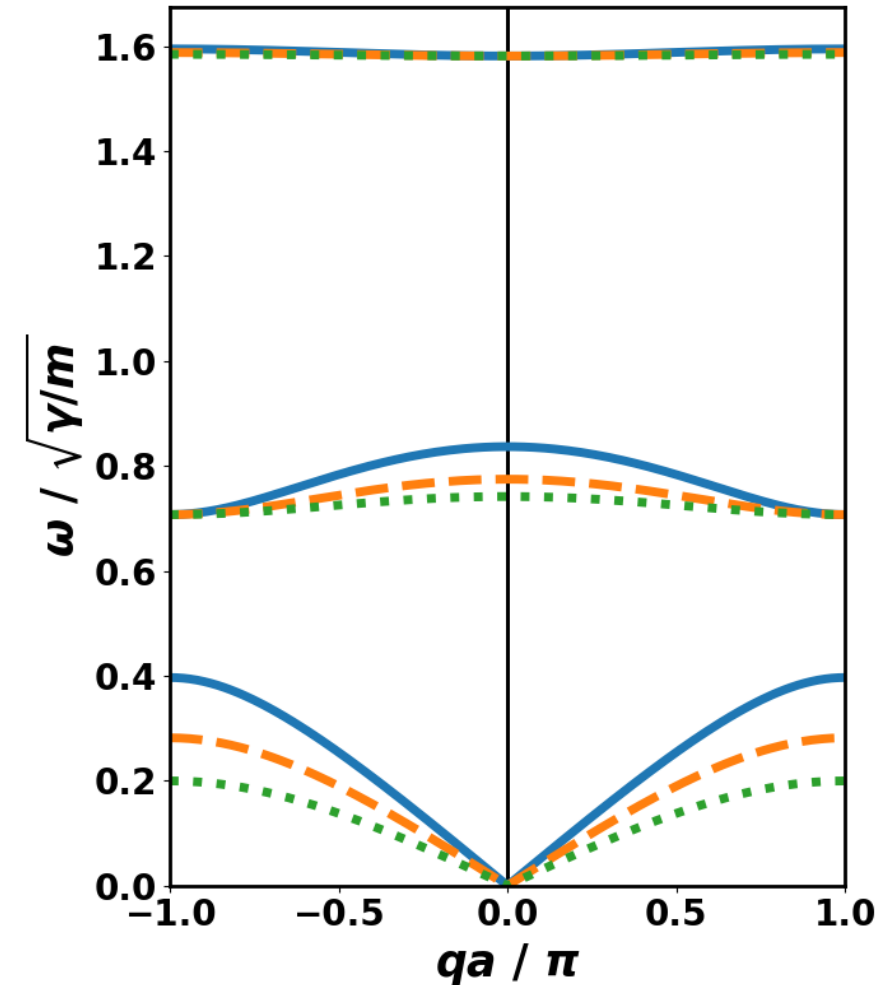
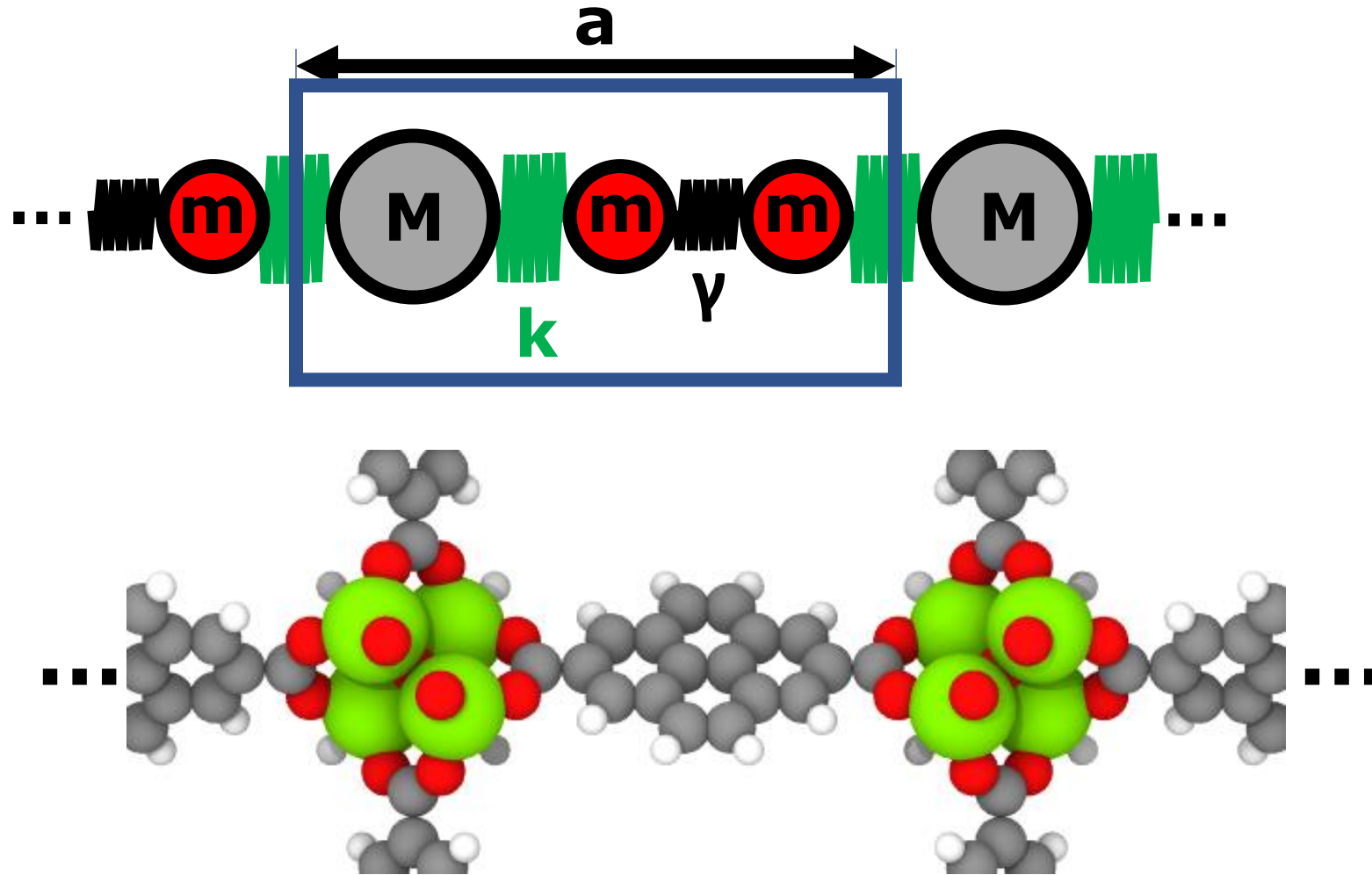
MOF-74

Raman intensity / a.u.

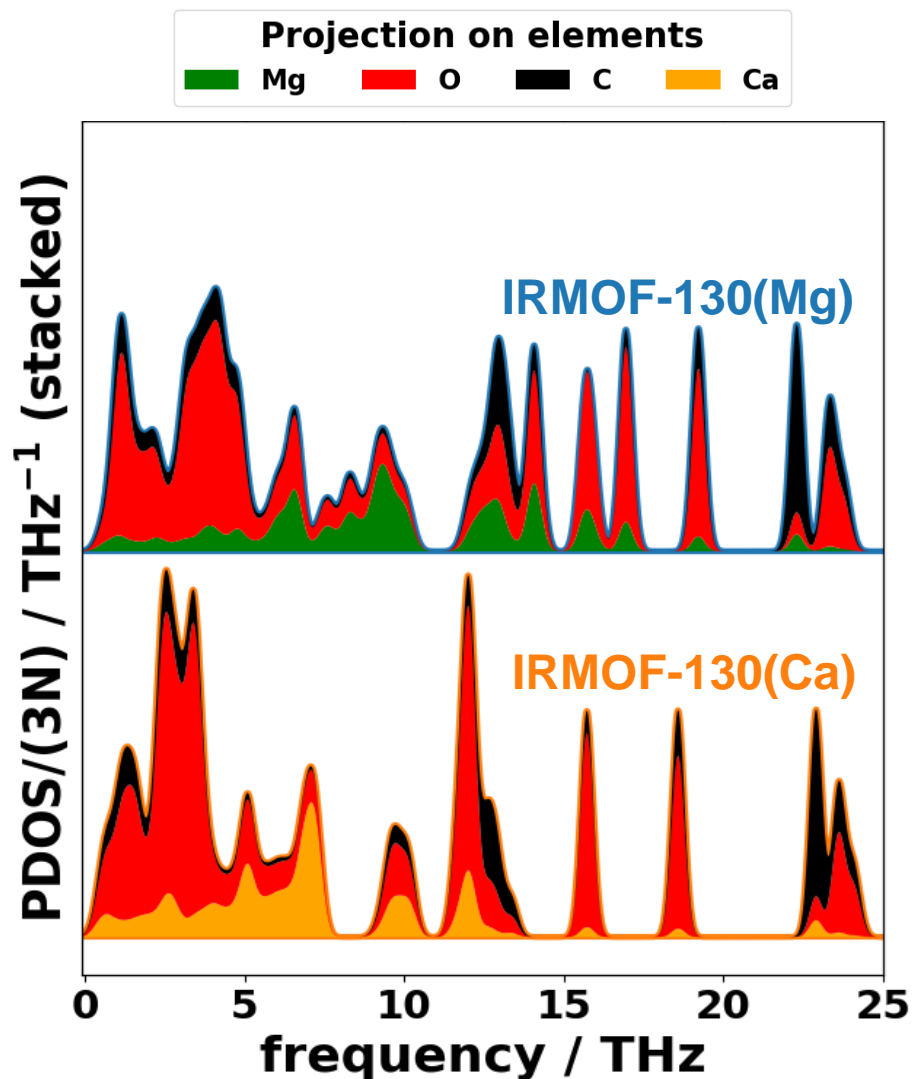
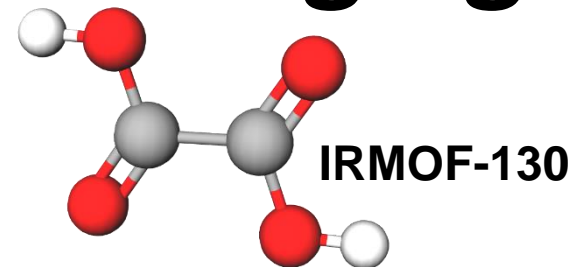


Kamencek, Wieser, Zojer et al., *J. Chem. Theory Comput.* **2020**, 16, 4, 2716–2735

Which Phonon Bands to Expect?



Projected Density of States: Exchanging Metals



Low frequency region

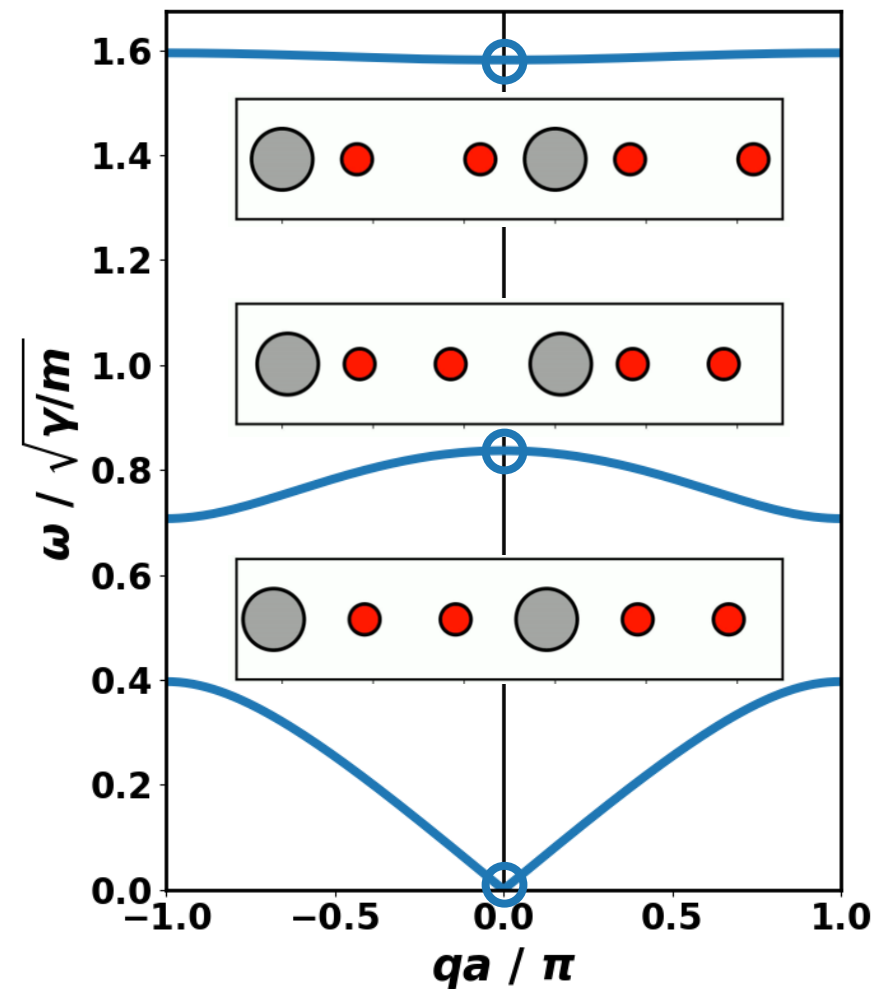
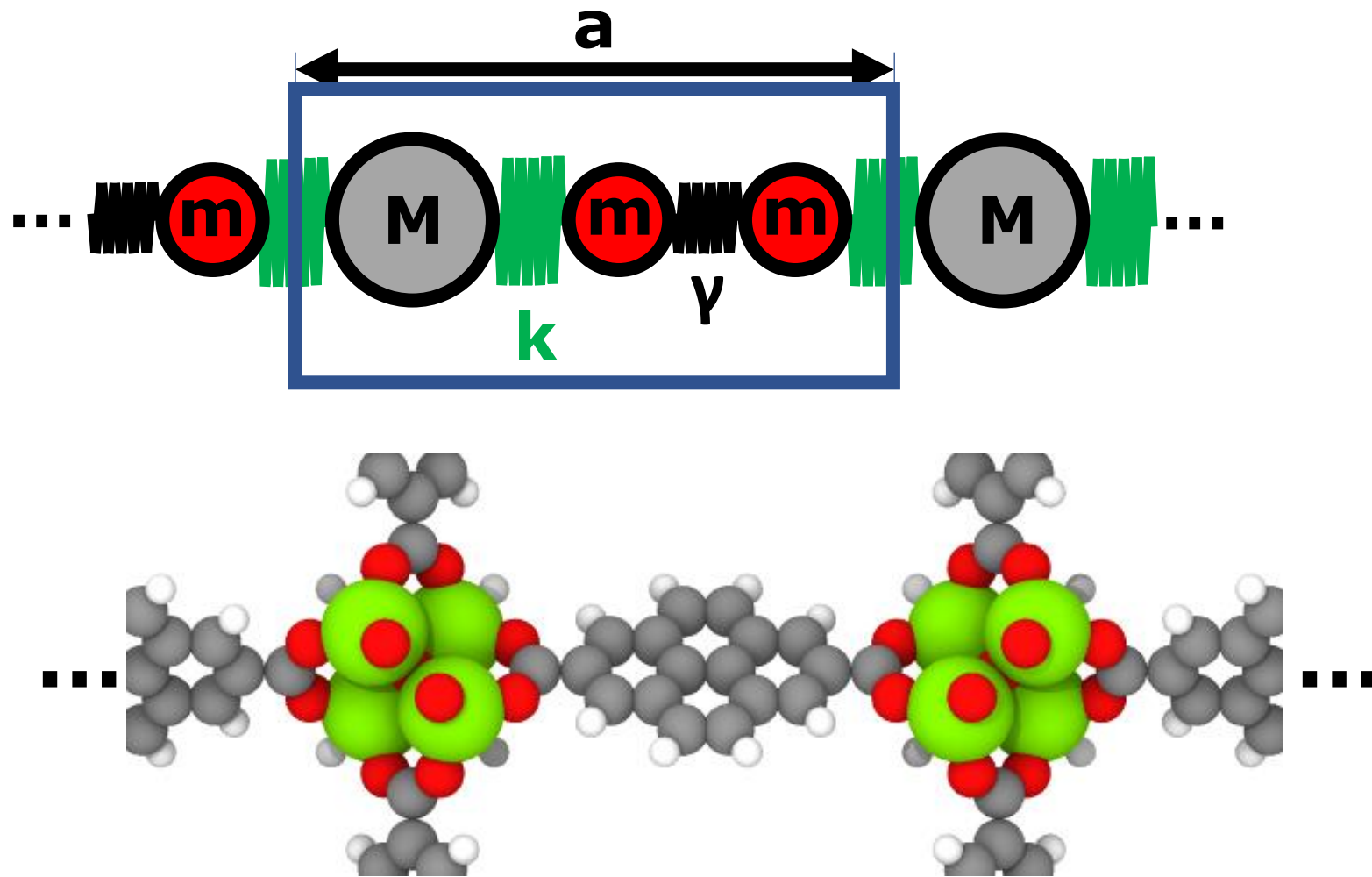
- dominated by O and Mg/Ca

Higher frequencies

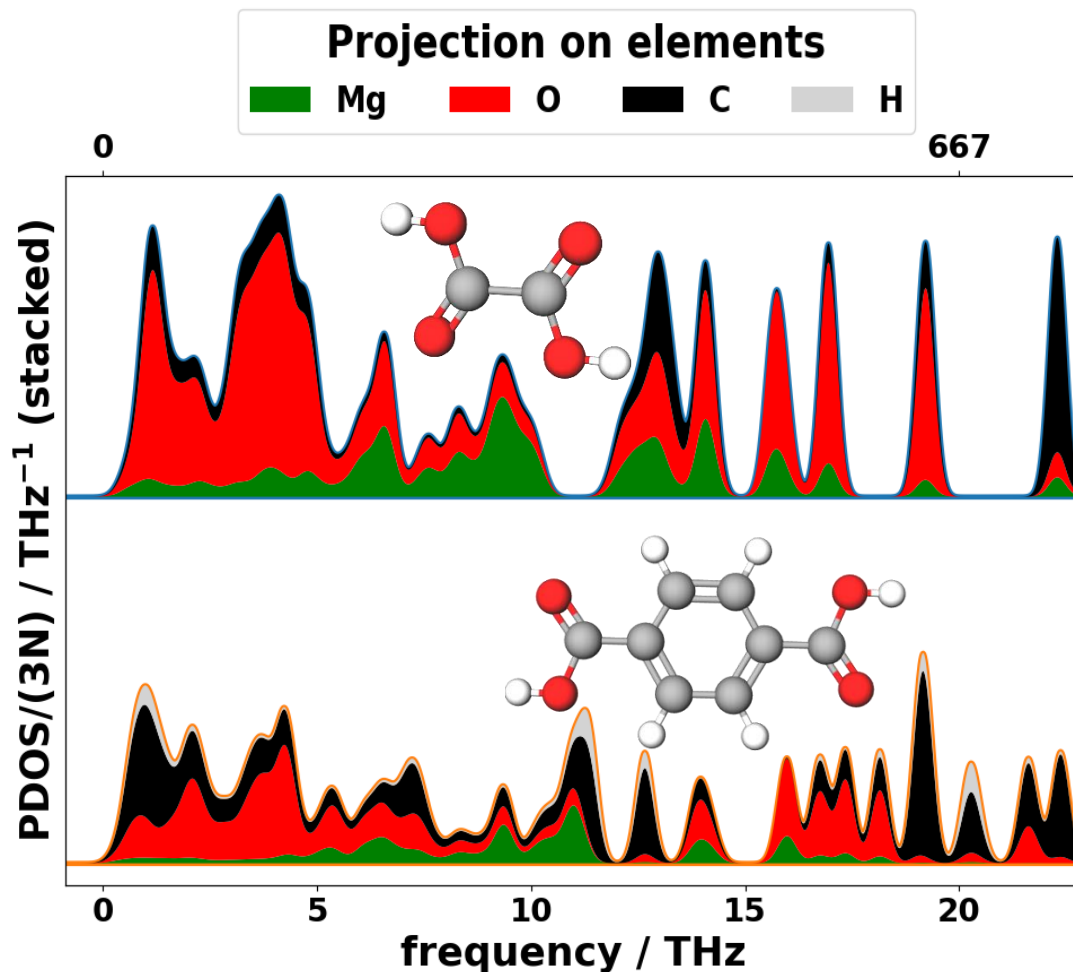
- C contributions
- Sharp peaks

Mg → Ca: Modes shifted to lower frequencies

Which Phonon Bands to Expect?

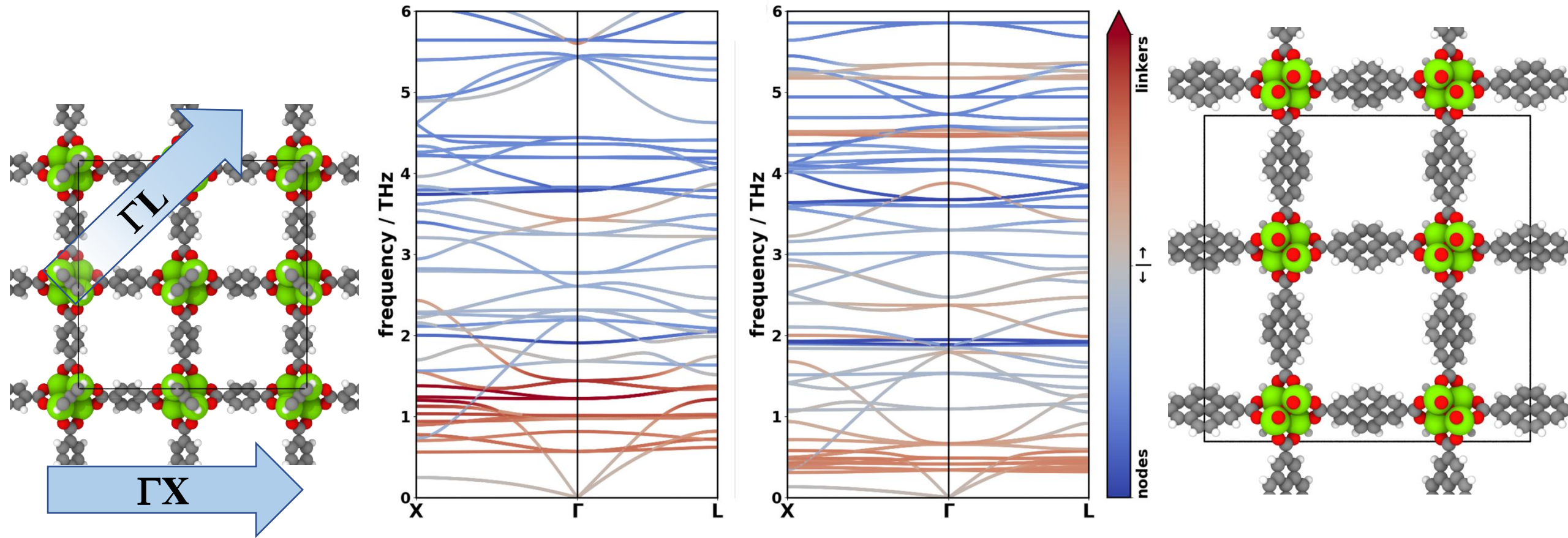


Projected Density of States: Exchanging Linkers

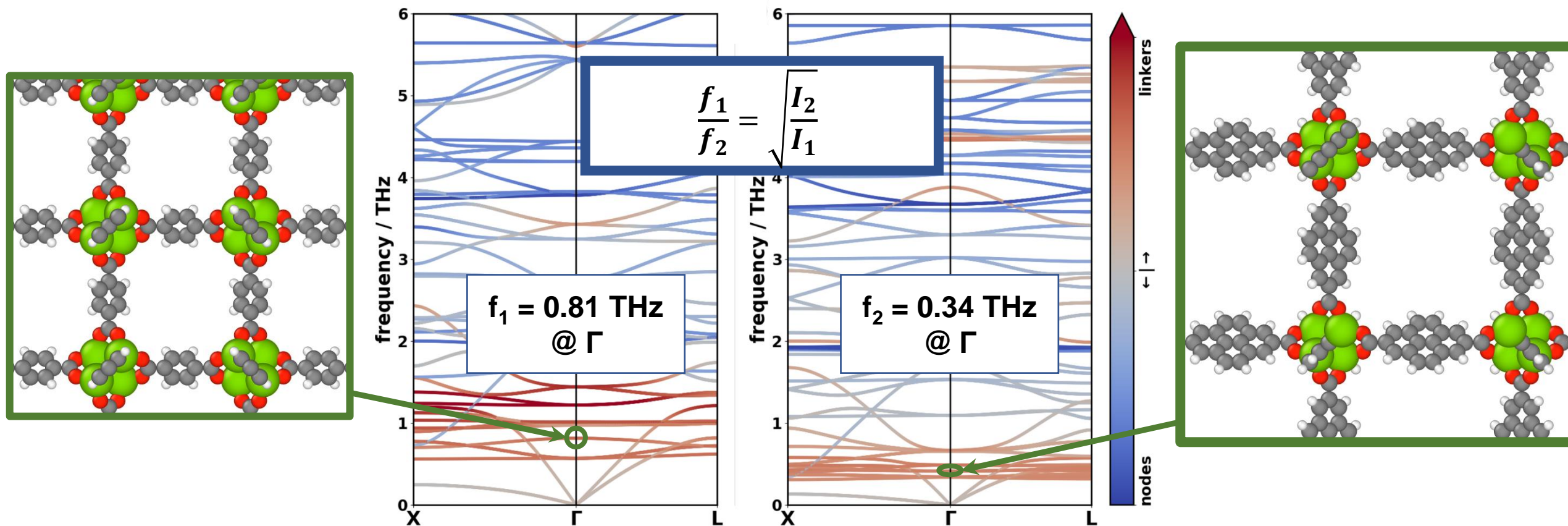


- more C contribution
- **low energy** motion (torsion, bending, rotation of linkers): **rigid unit modes**

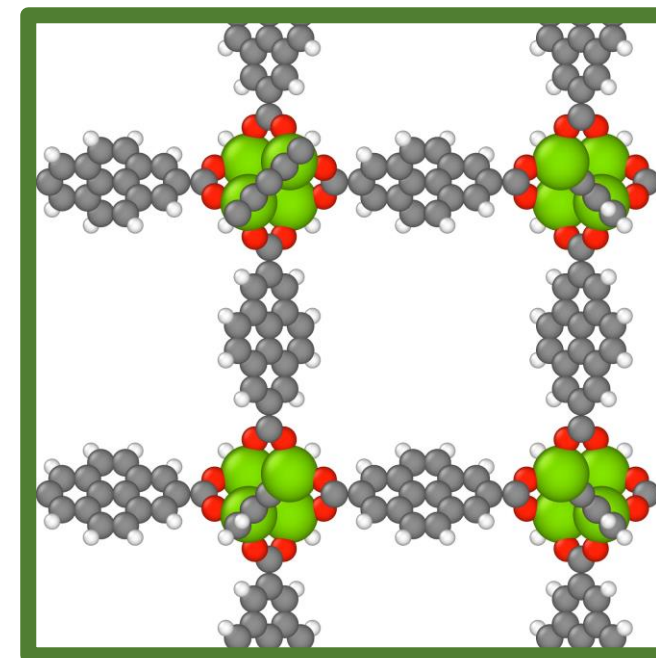
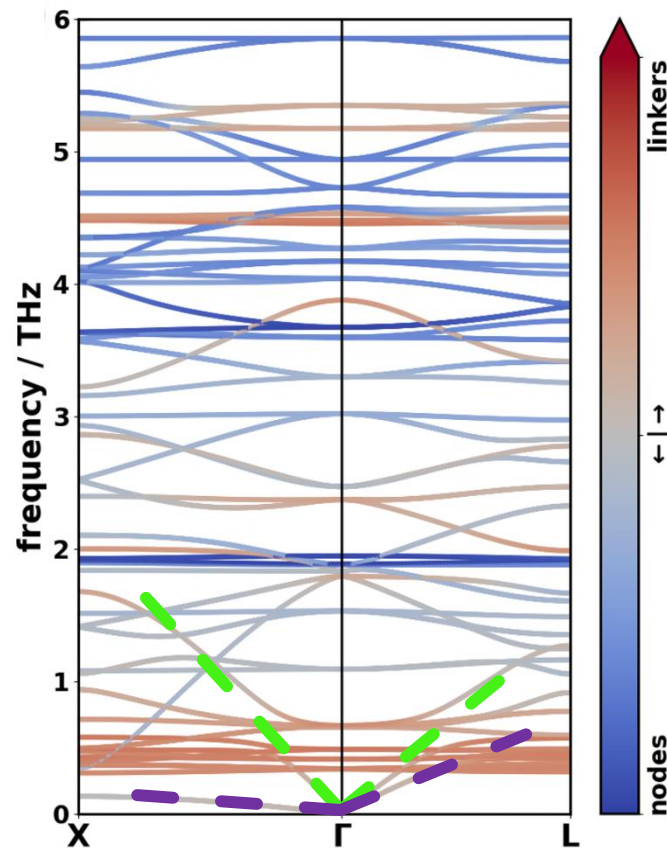
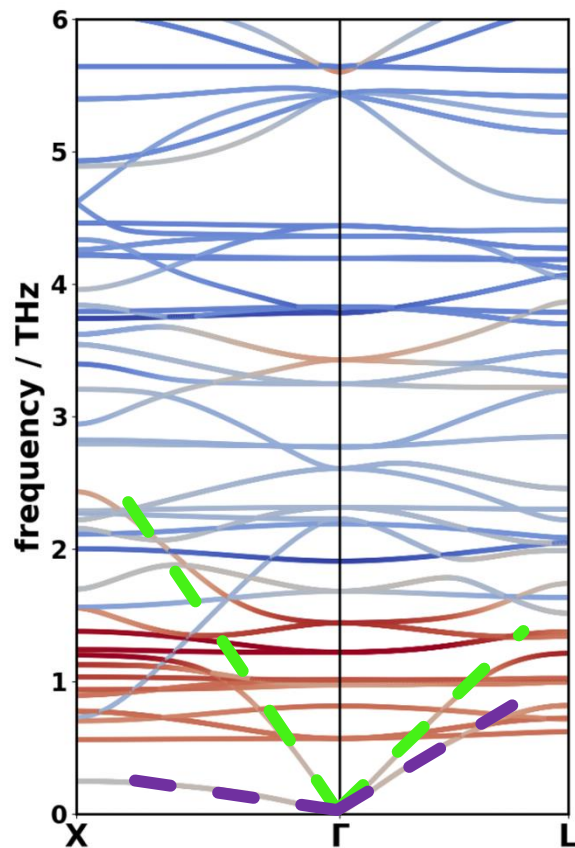
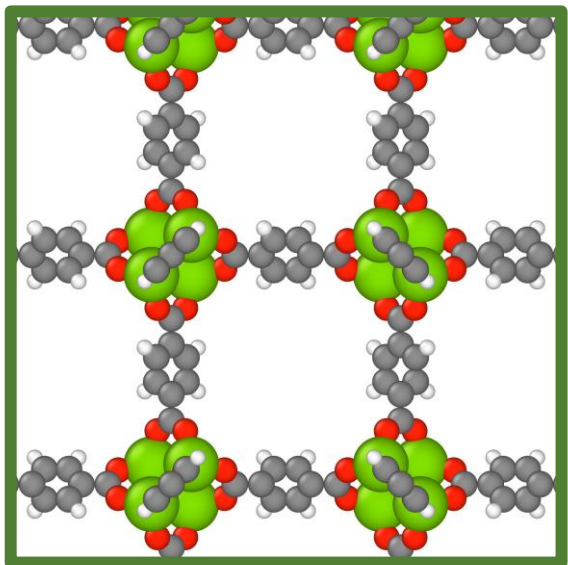
Increasing Complexity of the Linker: More Low-Energy Linker Modes



Increasing Complexity of the Linker: Quantitative Trends



Increasing Complexity of the Linker: Quantitative Trends

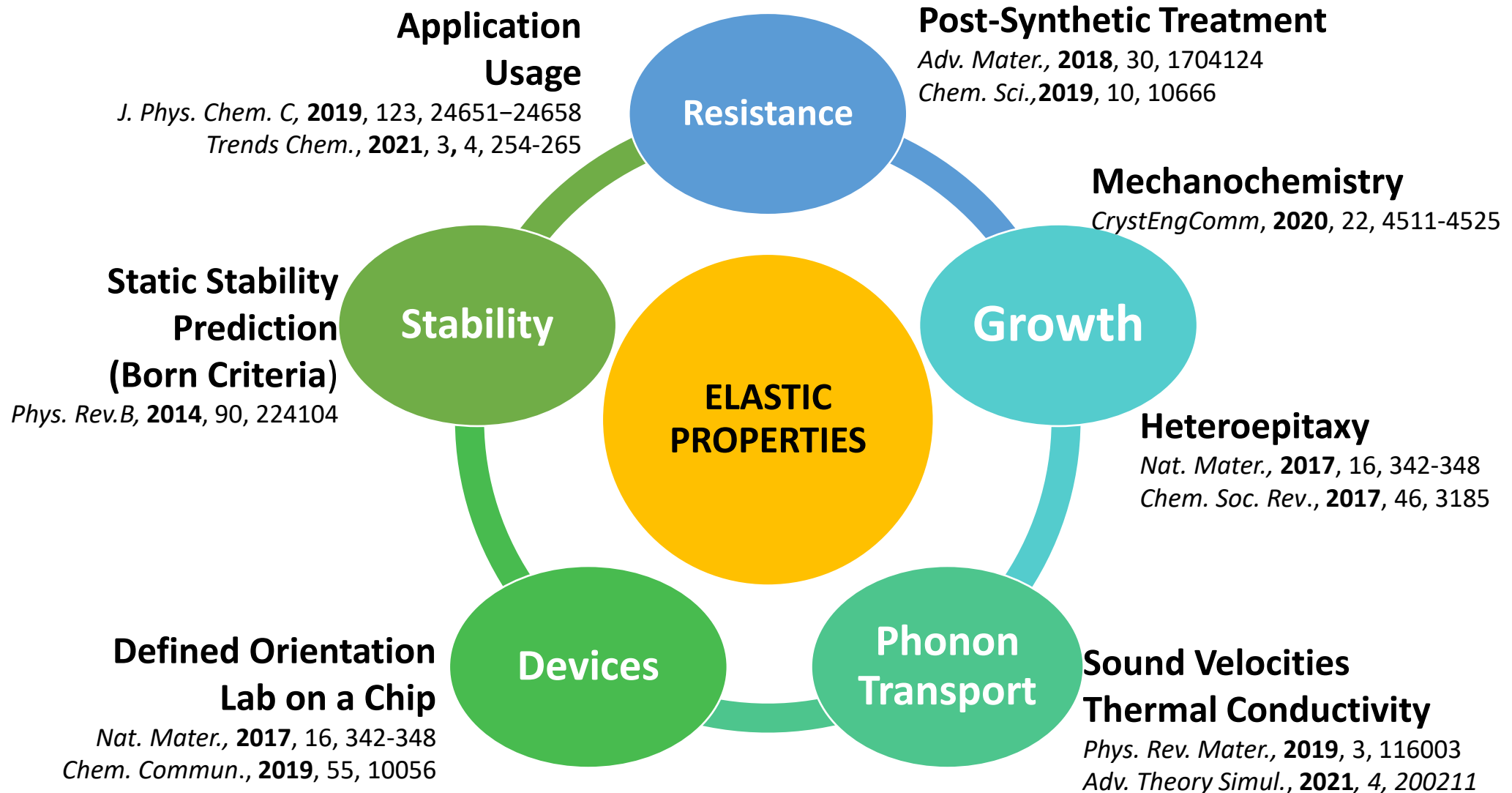


The background of the slide is a 3D visualization of a MOF-74 crystal structure. It consists of a dense packing of spheres. The spheres are colored in three ways: white, red, and green. The white spheres are the most numerous and form a continuous matrix. The red and green spheres are interspersed within this matrix, representing different types of atoms or functional groups within the metal-organic framework. The overall appearance is that of a complex, porous crystalline lattice.

Elastic Properties and Their Atomistic Origins in Variants of MOF-74

Kamencek and Zojer; *J. Phys. Chem. C* 2021, in press

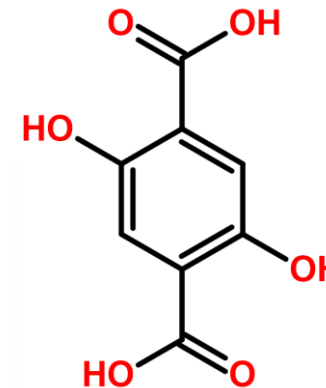
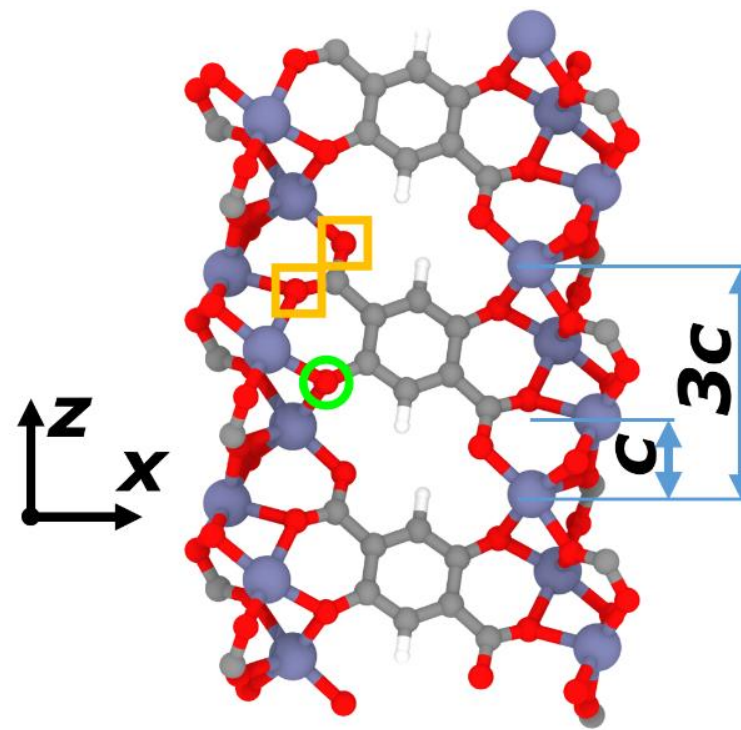
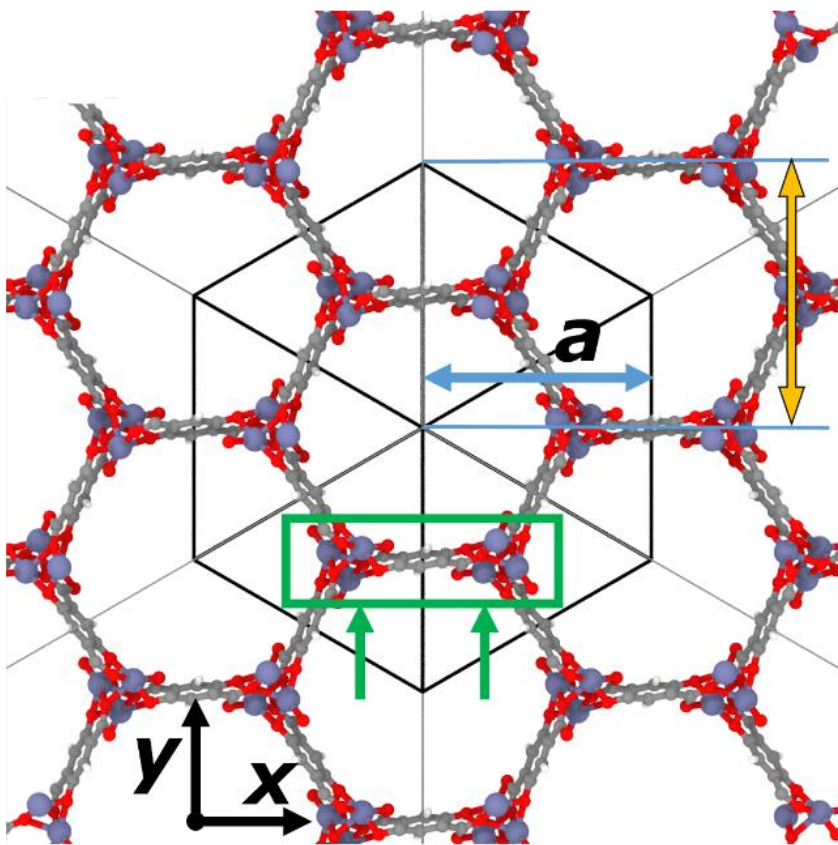
Why Elastic Properties?



MOF-74(Zn)

J. Am. Chem. Soc., 2005, **127**, 1504-1518
Science, 2012, **336**, 1018

Density Functional Theory: PBE/D3-BJ
 → Full Elastic Tensor

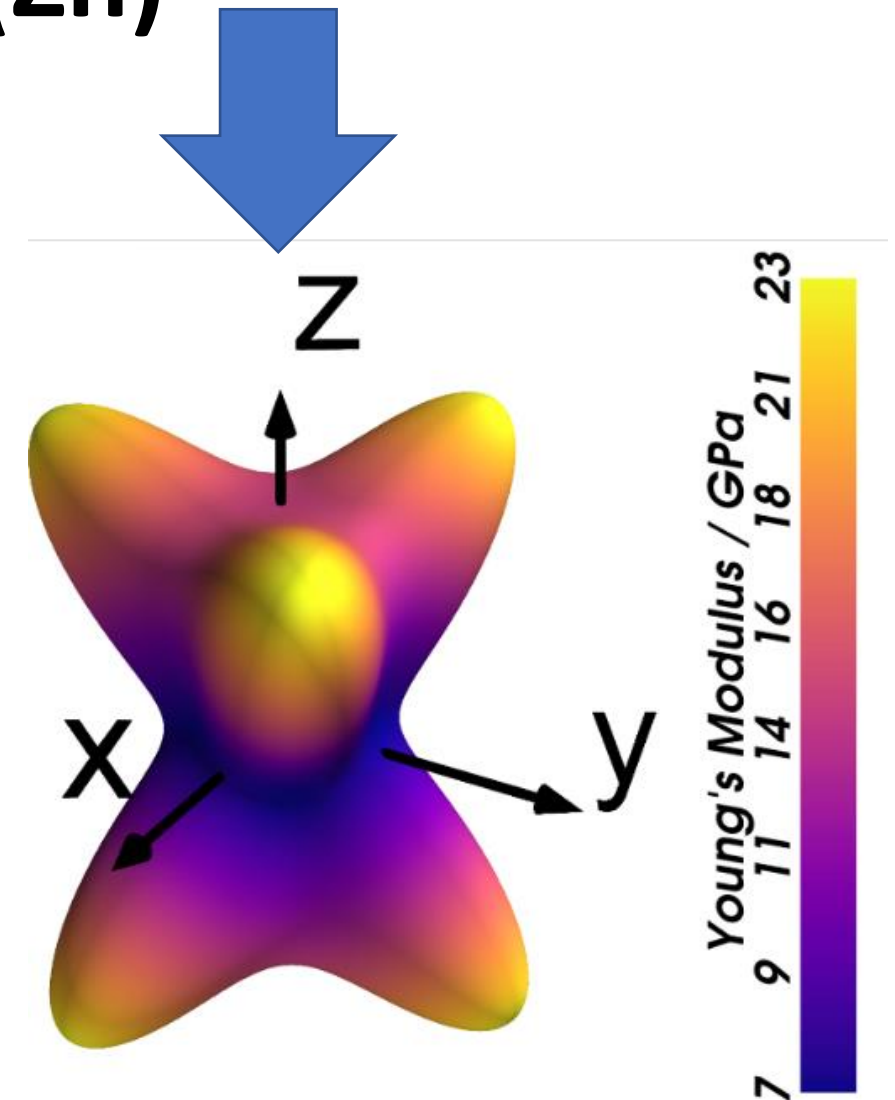


2 Lattice Parameters (a, c), rhombohedral Bravais lattice

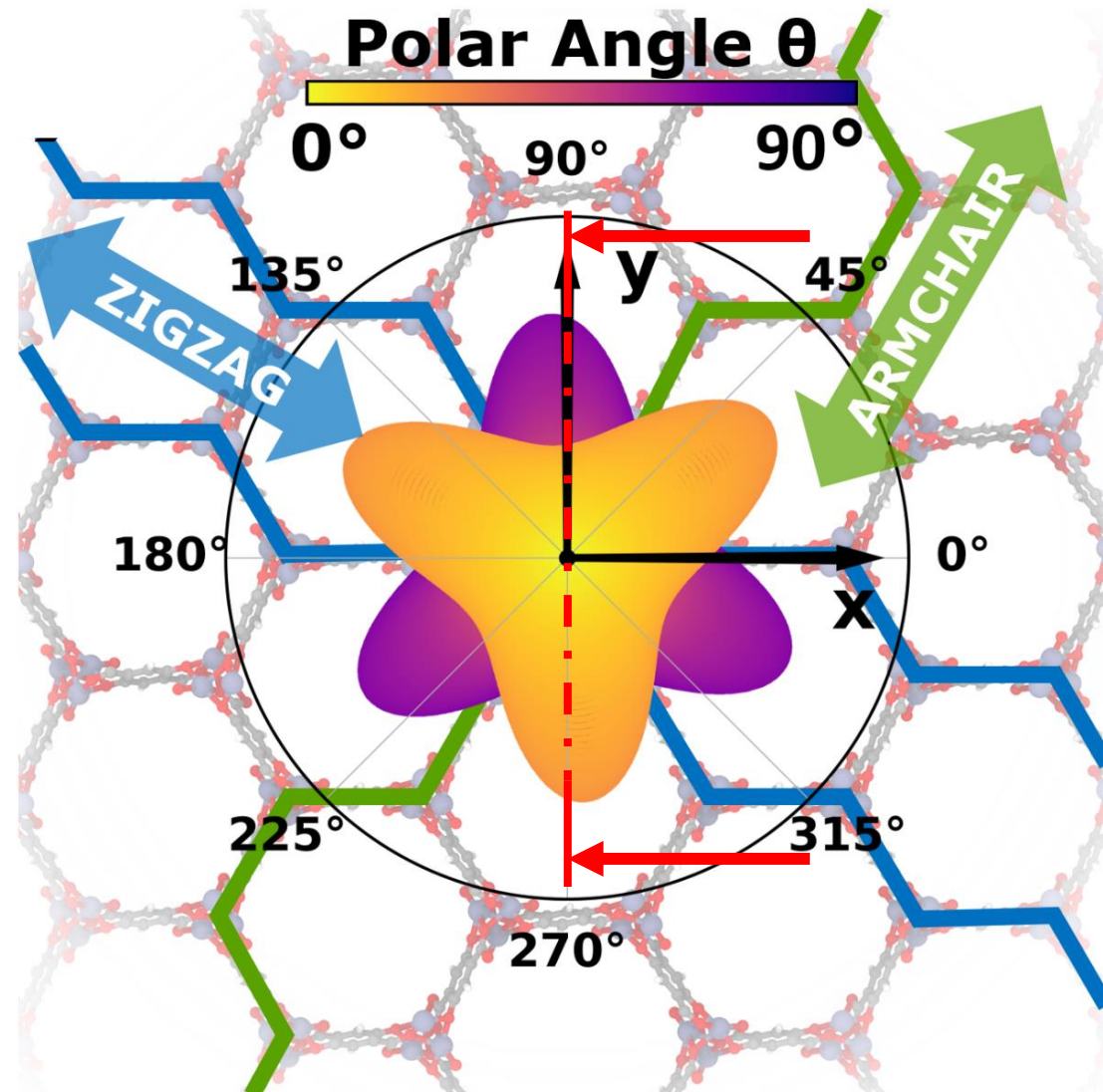
Young's Modulus in MOF-74(Zn)

= mechanical resistance to **uniaxial** strain

- Local minimum in **z-direction**
- Smaller in **xy-plane**
- 3 pronounced lobes ($\pm 48^\circ$ inclined with respect to **xy-plane**)

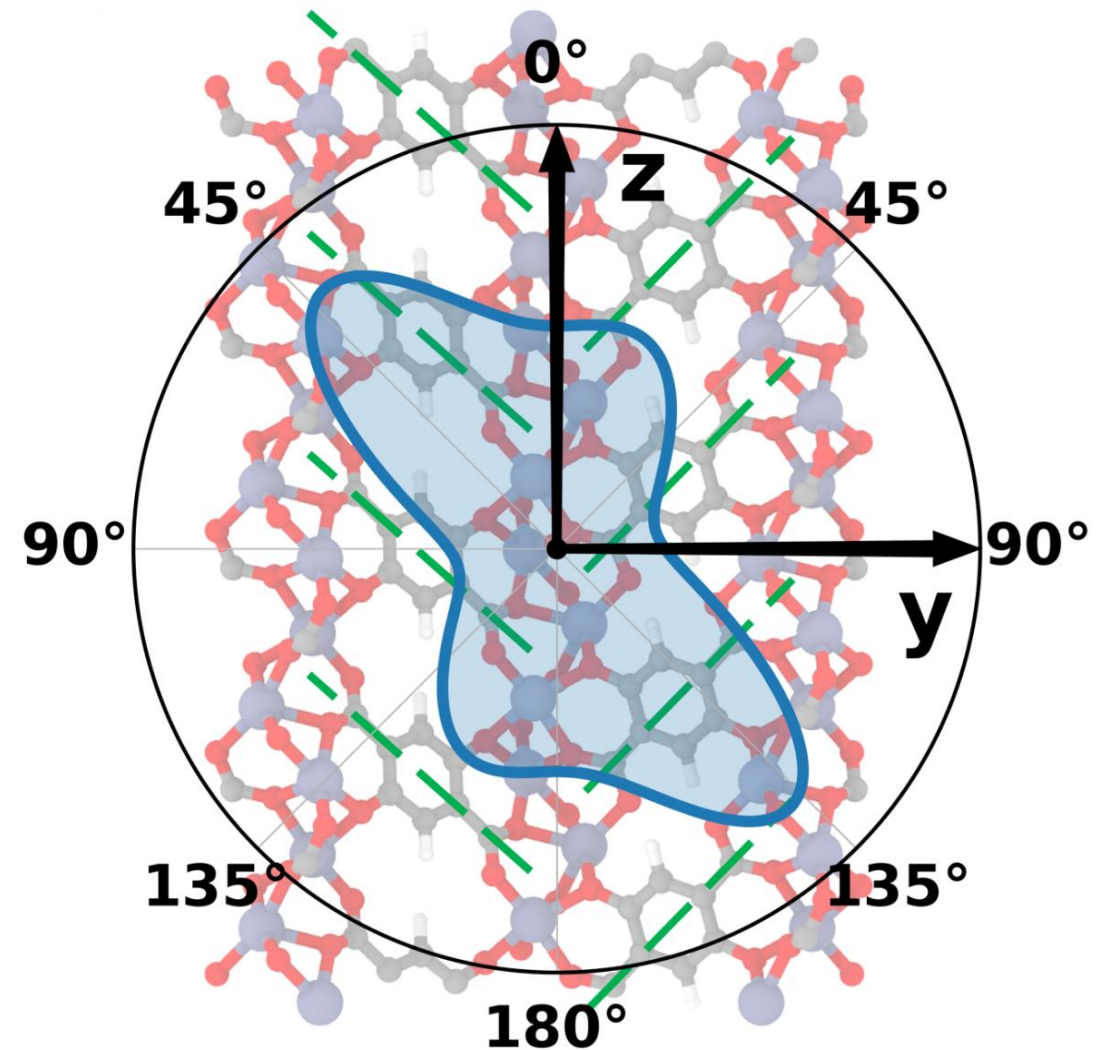


Young's Modulus in MOF-74(Zn) (top view)



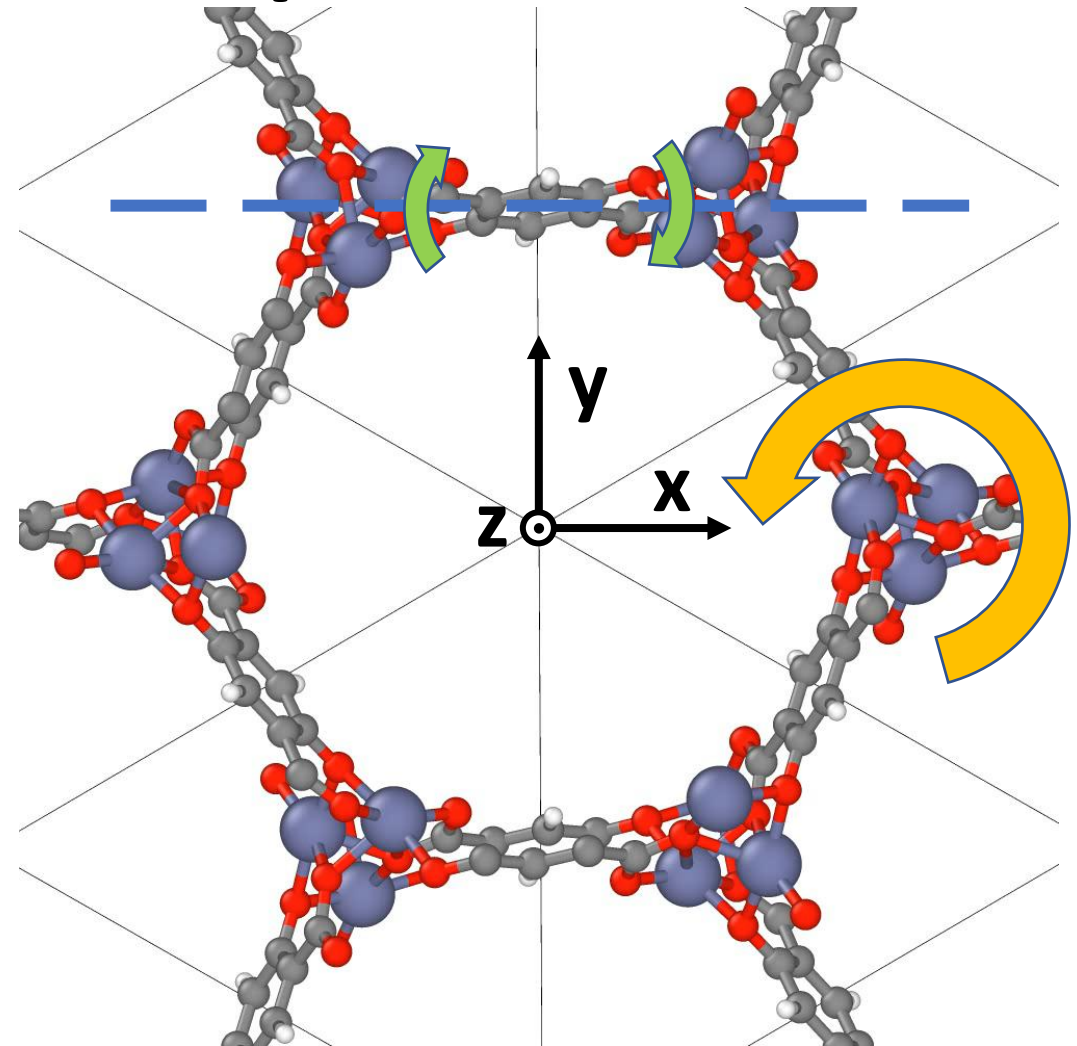
Young's Modulus in MOF-74(Zn) (side view)

- Maxima nearly aligned with the **long molecular axes** of the linkers (~5° difference)
- Linker backbone: strong covalent C-C bonds



Mechanisms at the Microscopic Level

- Applying compressive stress in **z-direction**
 - Shrinkage along channel
 - Lateral expansion
- Nodes grow laterally
- Nodes rotate
- Linker inclination increases



Structure-to-Property: Microscopic Changes to Macroscopic Properties

Adsorbates	Metal Ions	Linker Molecules
<ul style="list-style-type: none">• Water	<ul style="list-style-type: none">• Zn²⁺• Ca²⁺• Mg²⁺• Be²⁺• Mixtures	<ul style="list-style-type: none">• 1 Phenylene ring• 2 Phenylene rings• 3 Phenylene rings

Structure-to-Property: Microscopic Changes to Macroscopic Properties

Adsorbates

- Water

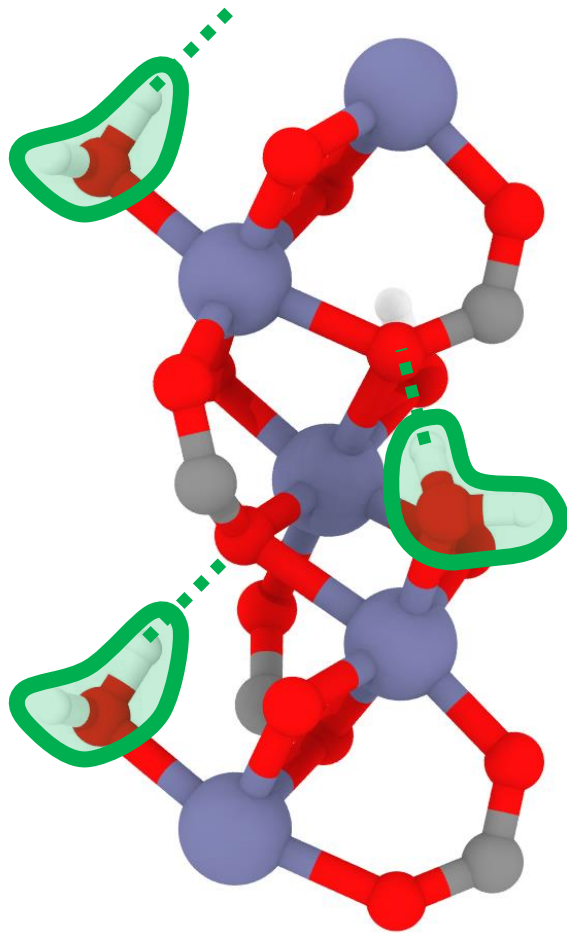
Metal Ions

- Zn^{2+}
- Ca^{2+}
- Mg^{2+}
- Be^{2+}
- Mixtures

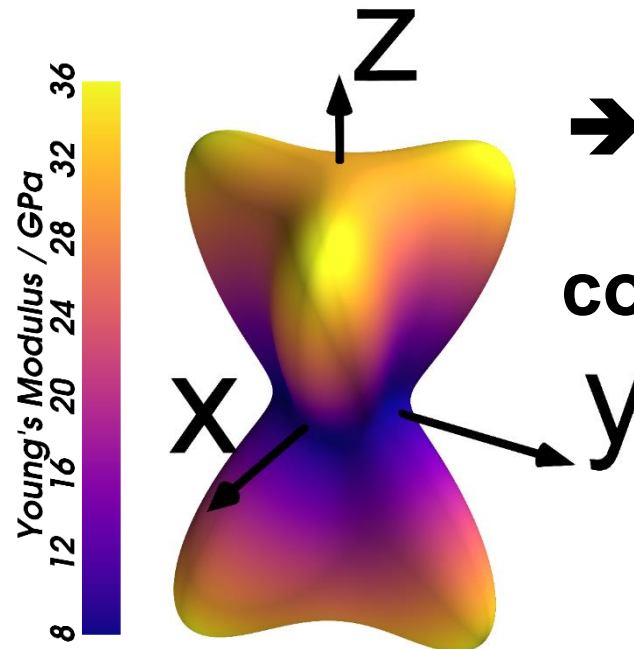
Linker Molecules

- 1 Phenylene ring
- 2 Phenylene rings
- 3 Phenylene rings

Effect of Adsorbed Water



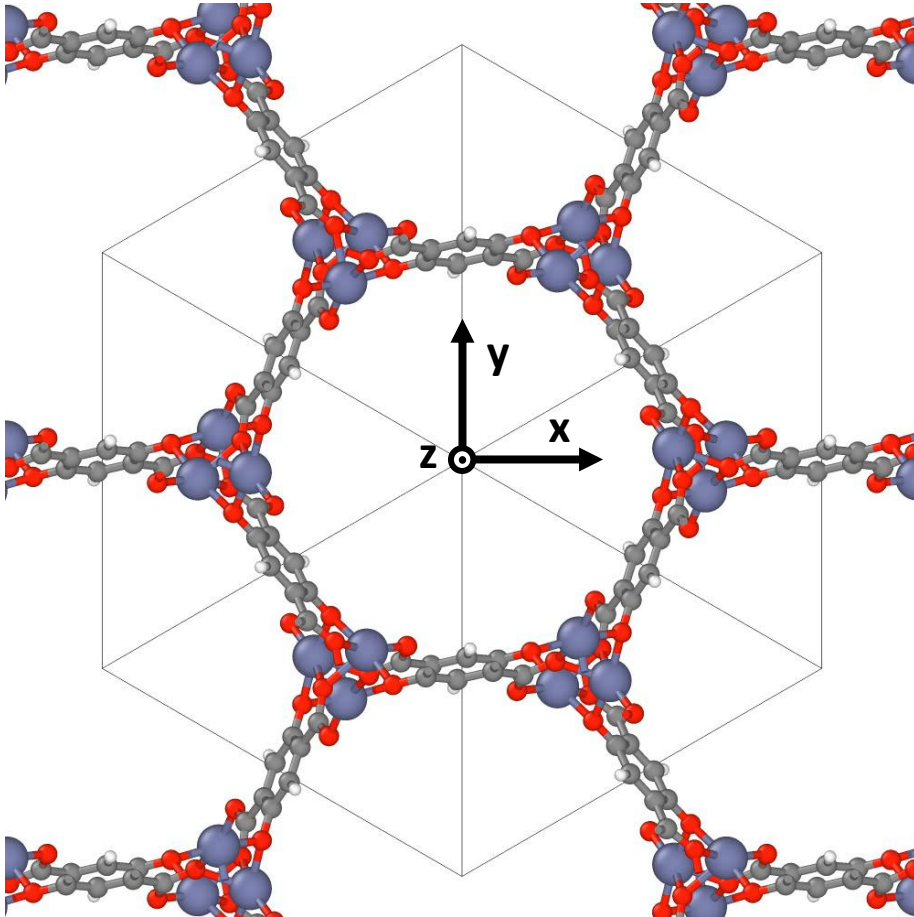
- Water at uncoordinated metal sites
- H-bridges
- Stabilize the node



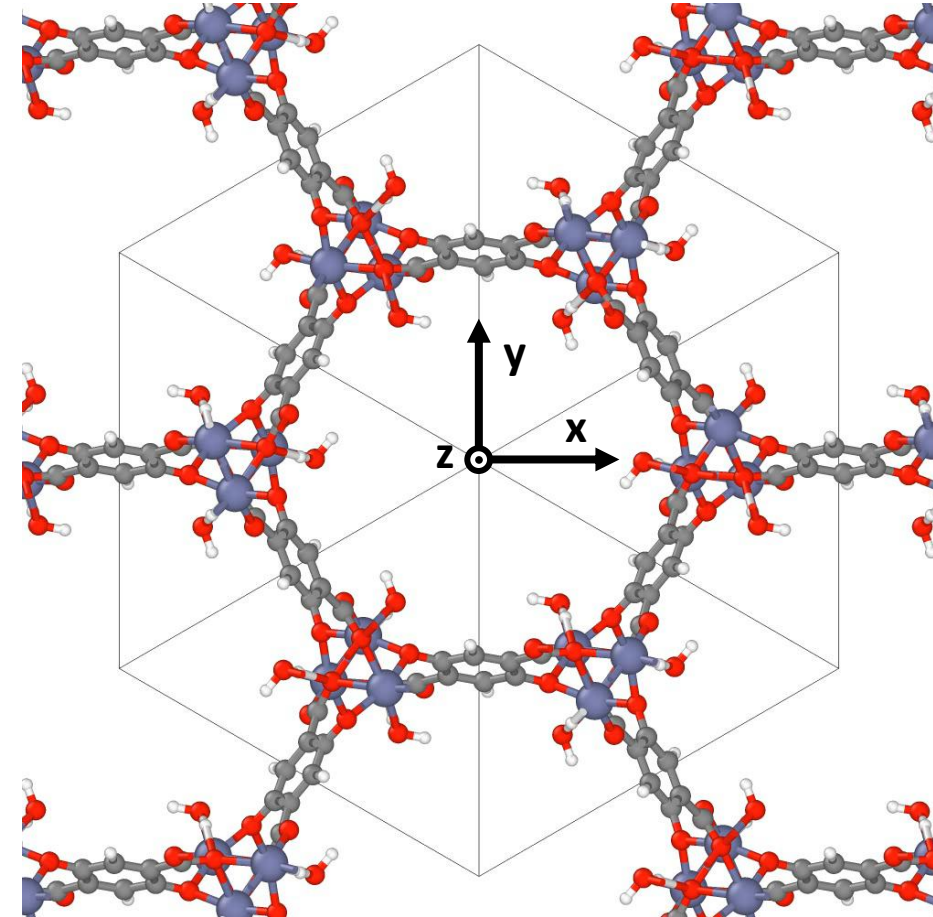
→ Young's Modulus in ***z***-
direction increases
compared to dehydrated
system

Hydrogen bridges make lateral expansion of nodes more difficult

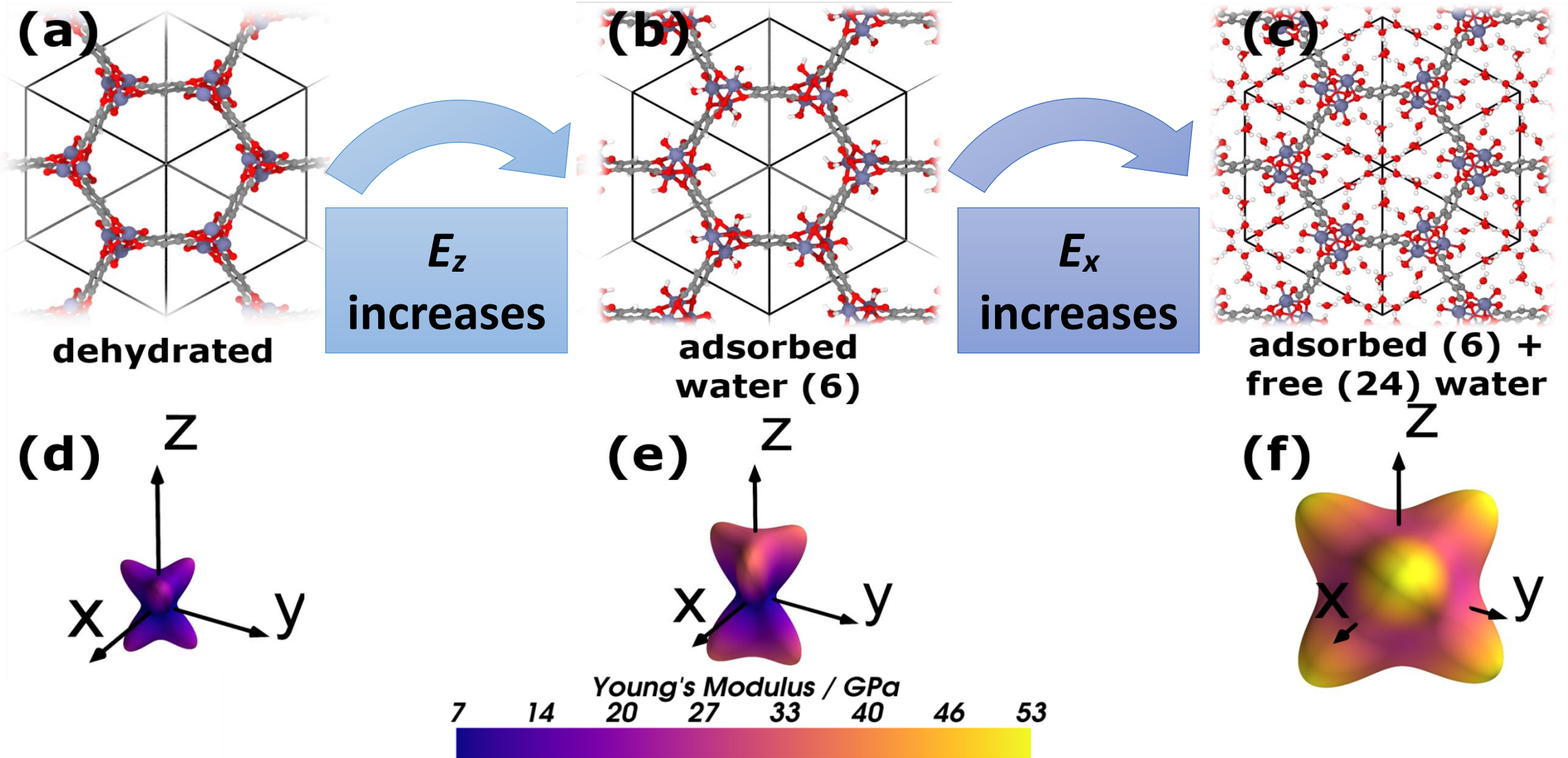
Dehydrated



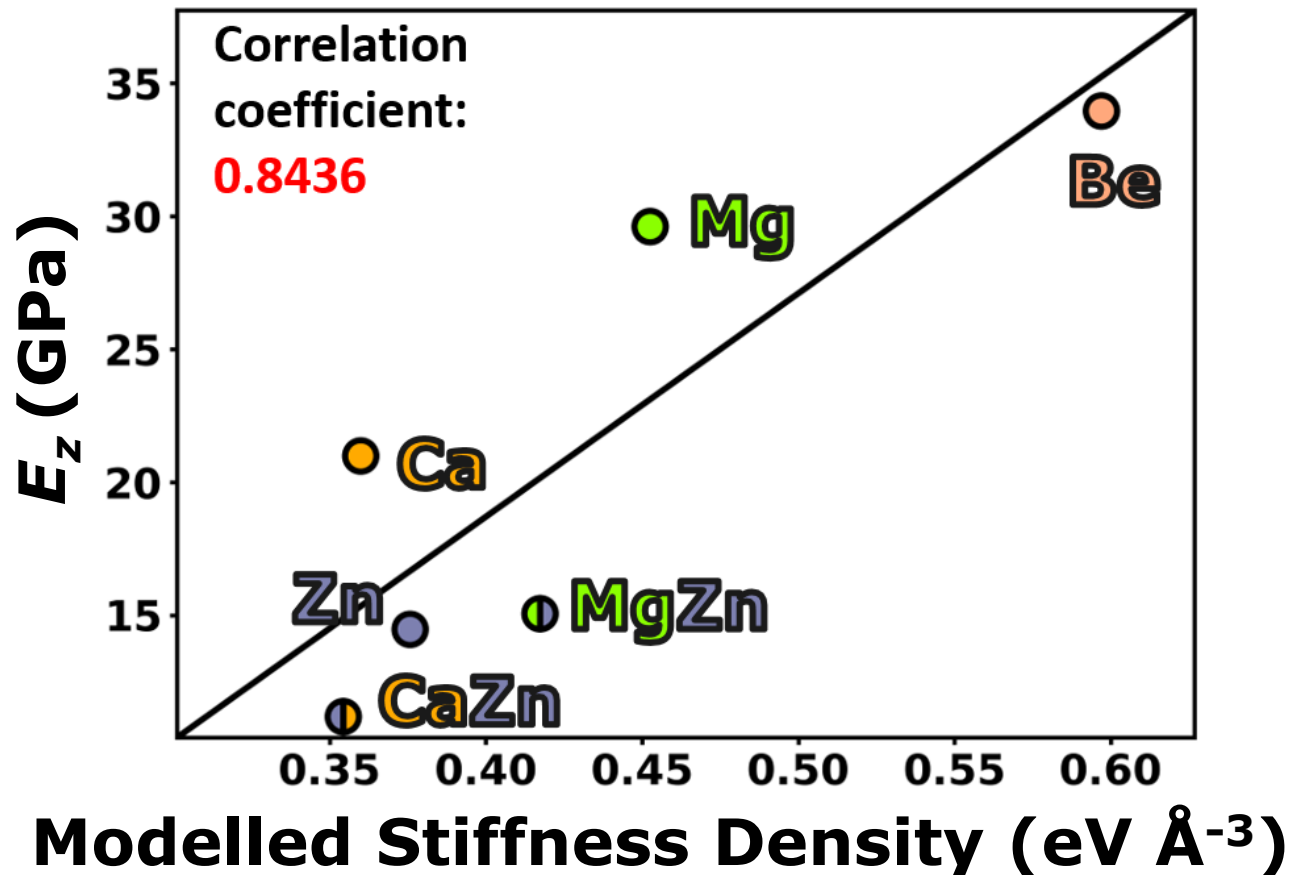
+6 H₂O per unit cell



Water increases E_z and after first layer also E_x



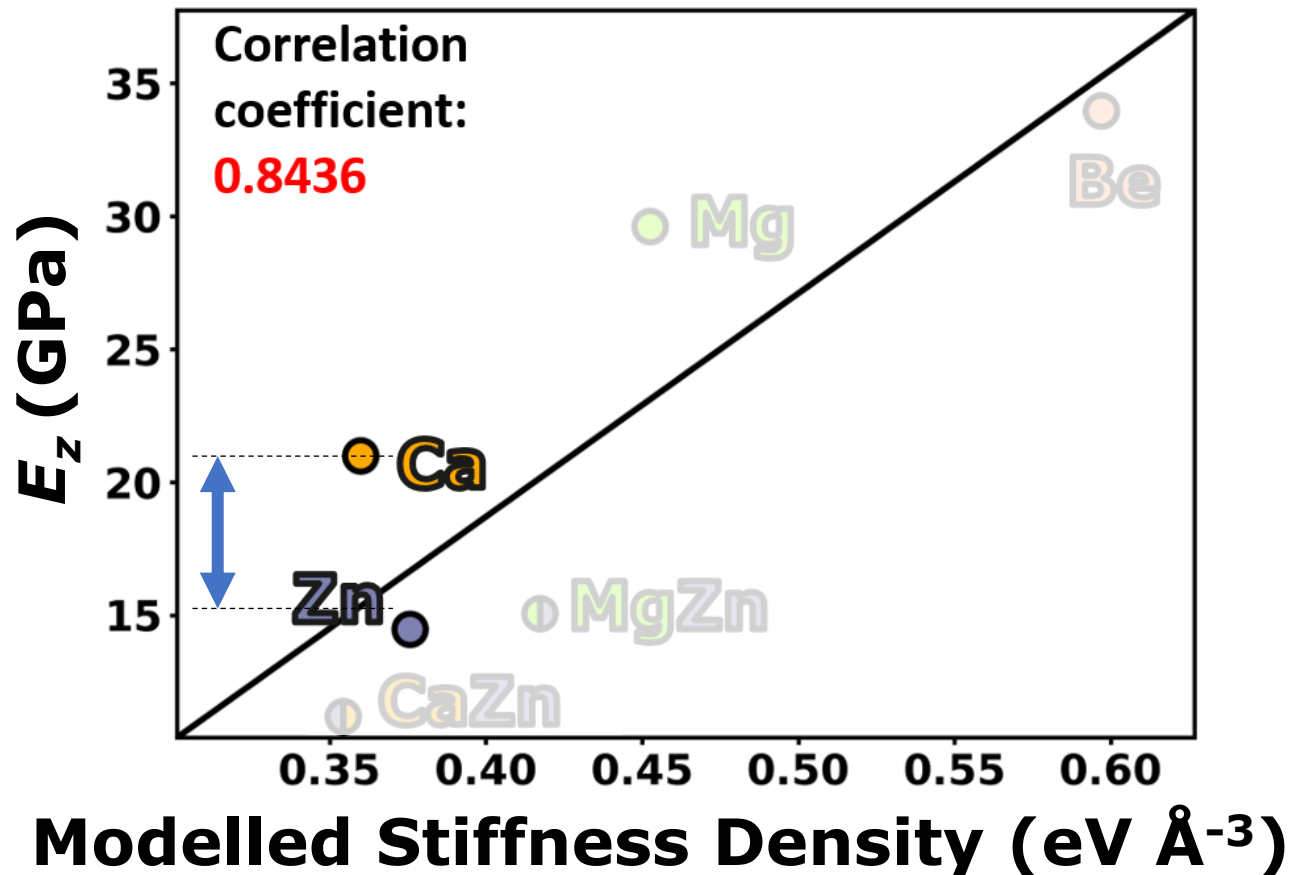
Metal Exchange: General Trends but Individual Deviations



Considering:

- **Porosity** (Lattice Constants)
- Metal-Oxygen **Interaction Strength** (Force Constants)

Metal Exchange: General Trends but Individual Deviations

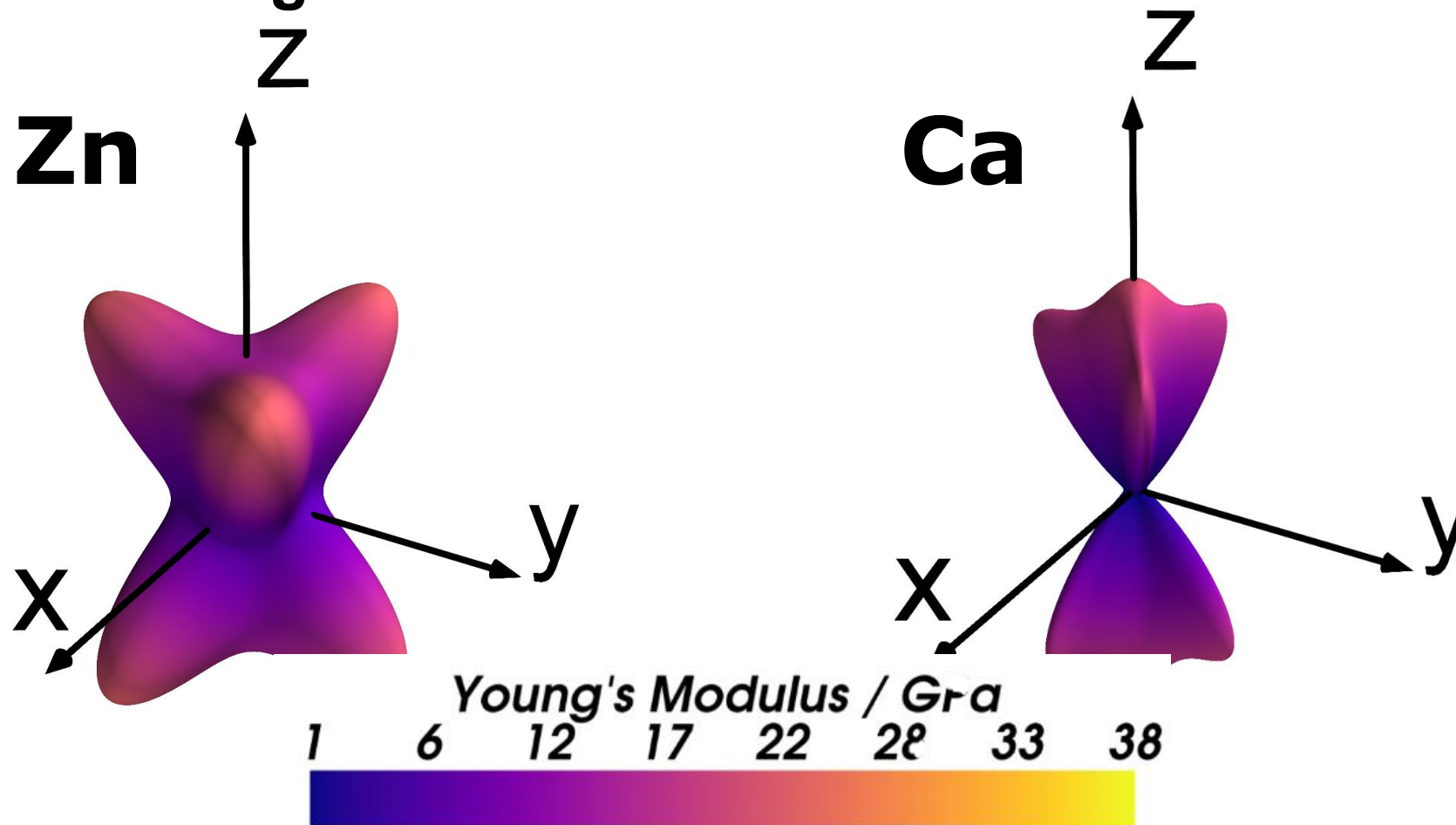


Considering:

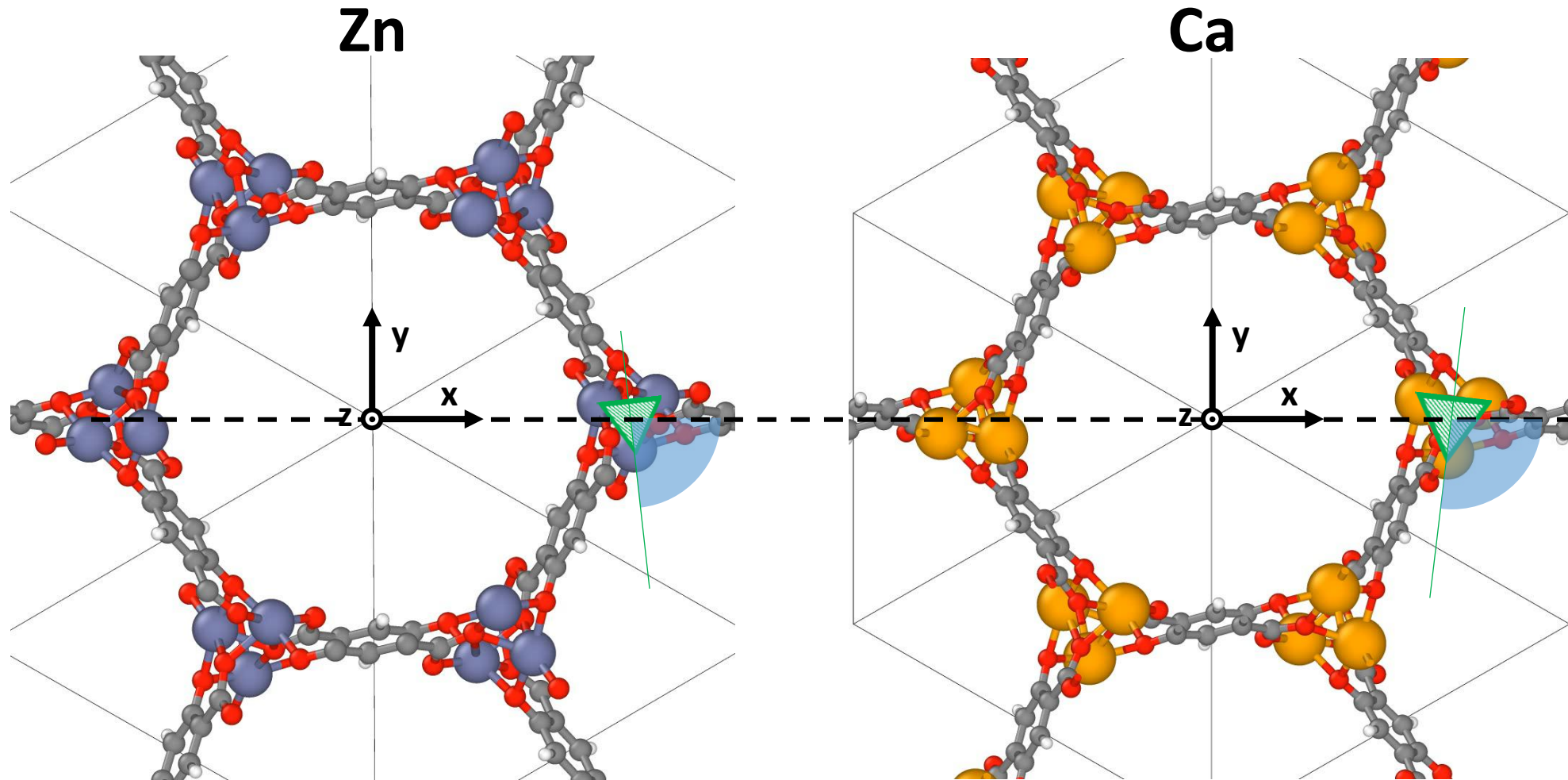
- **Porosity** (Lattice Constants)
- Metal-Oxygen **Interaction Strength** (Force Constants)

Metal Exchange: Young's Modulus Changes Also **Qualitatively**

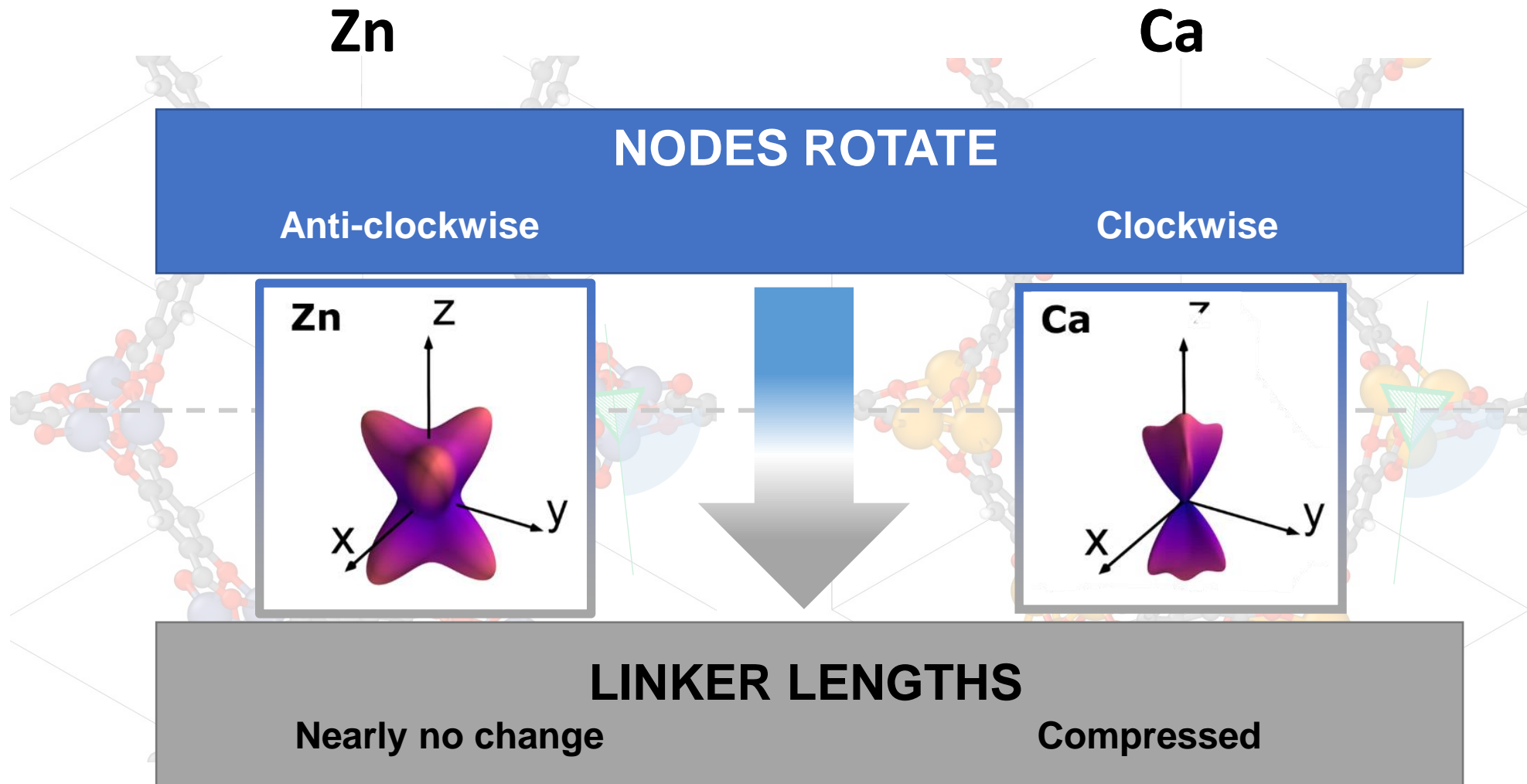
Ca: Young's modulus in z-direction is a **maximum**



Different Deformations upon z-Stress

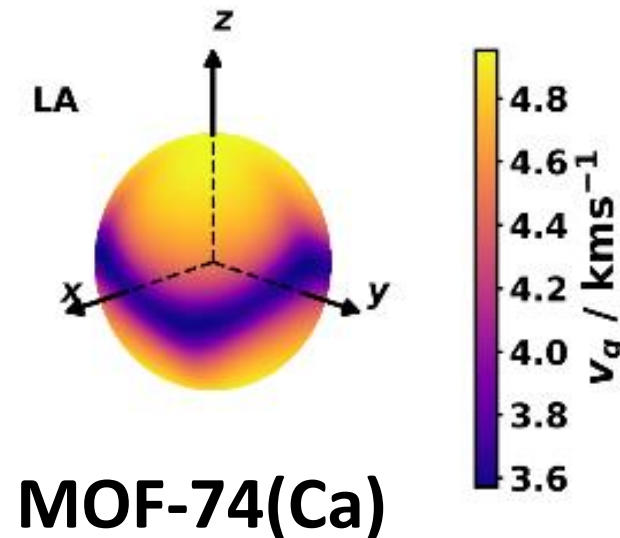
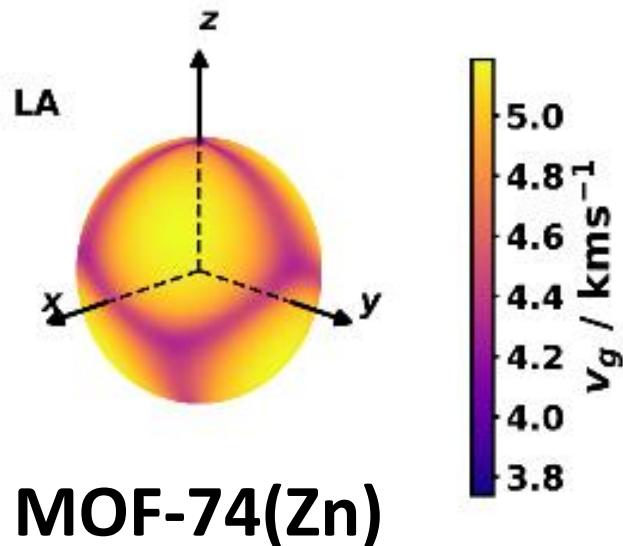


Different Deformations upon z-Stress



Sound Velocity Distributions Change

Elastic Properties \leftrightarrow Sound Velocities (Christoffel Equations)

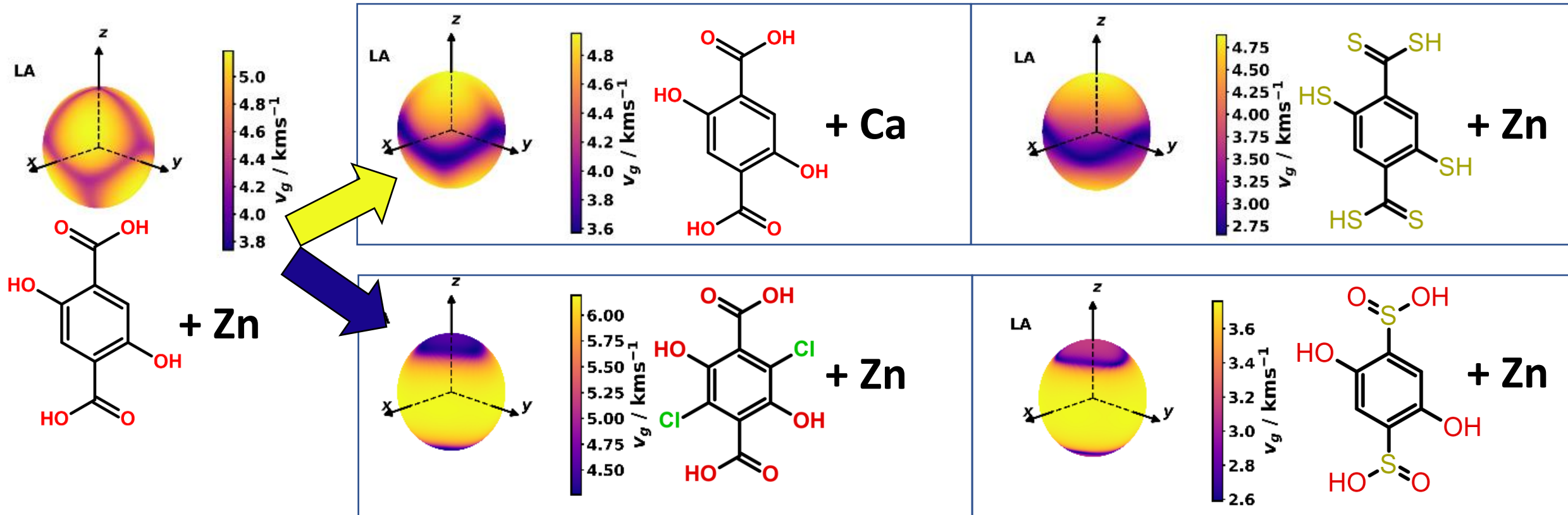


Longitudinal Acoustic Sound Velocities (Long Wavelength Limit)

Manipulation of Sound Velocity Distributions

Longitudinal Acoustic Sound Velocities (Long Wavelength Limit)

Elastic Properties \leftrightarrow Sound Velocities (Christoffel Equations)



A 3D ball-and-stick model of the MOF-74 structure, showing a complex, porous framework of atoms. The atoms are represented by spheres of different colors: red, white, and purple. The structure is highly symmetric and features large, interconnected pores.

Thermal Expansion and Anharmonic Properties in MOF-74

Kamencek and Zojer; *in preparation*

Thermal Expansion and Anharmonic Properties in MOF-74

Goals and Motivation

Institute of Solid State Physics and Institute of Physical and Theoretical Chemistry

ASP
DFT: PBE/D3-BJ

- Thermal expansion is an **anharmonic** effect
- Thermal expansion → **thermal mismatch (heteroepitaxy)**
- Origin of thermal expansion in MOFs **not fully clear** (different hypotheses)
- High level of insight: **Grüneisen theory of thermal expansion**

DocDay 09-2021

Grüneisen Theory of Thermal Expansion

Institute of Solid State Physics and Institute of Physical and Theoretical Chemistry

Cubic Crystals

$$\alpha_V = \frac{C_V \langle \gamma \rangle}{V B} \quad \gamma_\lambda = -\frac{V}{\omega_\lambda} \frac{d\omega_\lambda}{dV} \quad \text{Anharmonic Effect!}$$

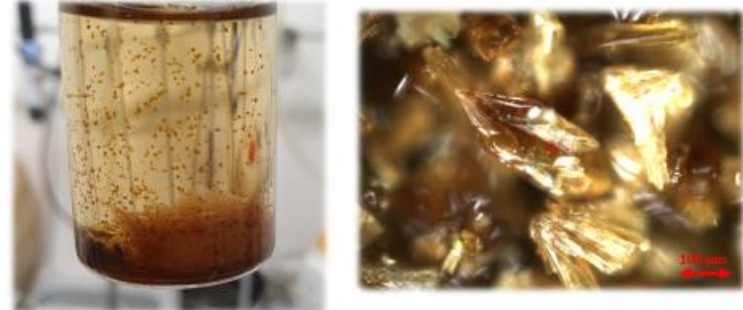
Non-Cubic Crystals

$$\alpha_{ij} = \frac{C_V}{V} S_{ijkl} \langle \gamma_{kl} \rangle \quad \gamma_{kl}^\lambda = -\frac{1}{\omega_\lambda} \frac{d\omega_\lambda}{d\varepsilon_{kl}} \quad \text{Anharmonic Effect!}$$

DocDay 09-2021

Real-Life MOF-74(Zn)

Institute of Solid State Physics and Institute of Physical and Theoretical Chemistry



DocDay 09-2021

Large mass mismatch between metals and linkers leads to complex phonon properties

Heavier metals shift phonons to lower frequencies

More complex linkers introduce more low-lying linker deformation modes

Inter-system trends of Γ -phonon frequencies can be observed based on classical arguments

Phys. Rev. Mater.
2019, 3, 116003

Maxima of Young's Modulus in MOF-74(Zn) are found along the (inclined) zigzag directions

(The first layer of) water increases mostly Young's Modulus along the pore

The trend of ion exchange can be estimated with a simple spring model

Individual deviations from the trends must be analysed in detail exploiting atomic deformations

J. Phys. Chem. C
2021, in press

Thermal expansion in MOF-74(Zn) is extremely low (XRD and Grüneisen Theory)

Most relevant contributions to the thermal expansion can be found for phonons < 3 THz

The acoustic phonons (especially in Γ_F and Γ_P) show the most pronounced anharmonicities

In preparation

Acknowledgements

- Supervisors: Paolo Falcaro and Egbert Zojer
- Falcaro Group @ IPTC and
- Advanced Modelling Team @ IF
- Austrian Academy of Sciences: DOC Fellowship
- Lead Project (LP-03) Porous Materials @ Work
- Computational Resources: VSC, High-Performance Computing TU Graz (dCluster, aCluster)

ÖAW

ÖSTERREICHISCHE
AKADEMIE DER
WISSENSCHAFTEN

Grant No. 25783

VIENNA
SCIENTIFIC
CLUSTER

TU
Graz

Goals and Motivation



DFT: PBE/D3-BJ

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- Thermal expansion → **thermal mismatch (heteroepitaxy)**
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(different hypotheses)
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Grüneisen Theory of Thermal Expansion

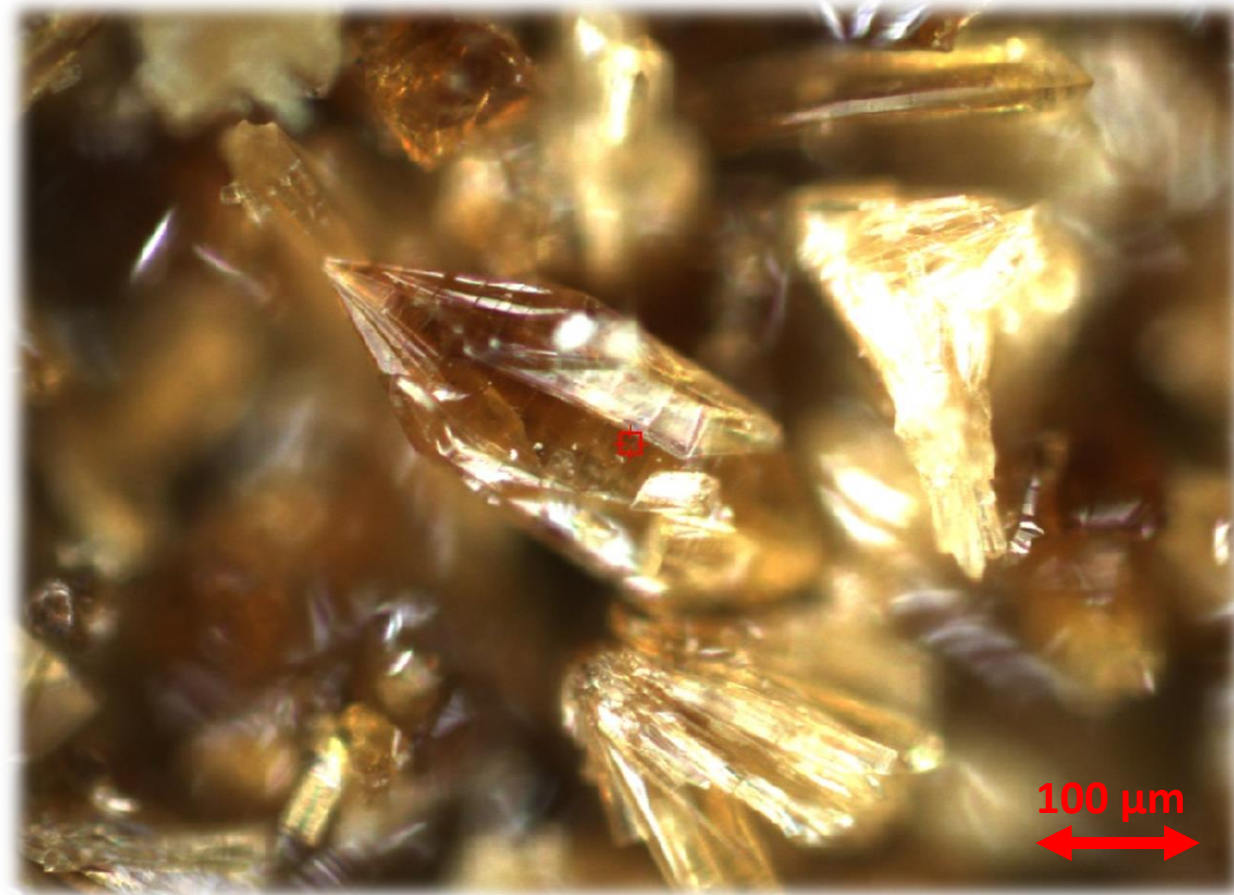
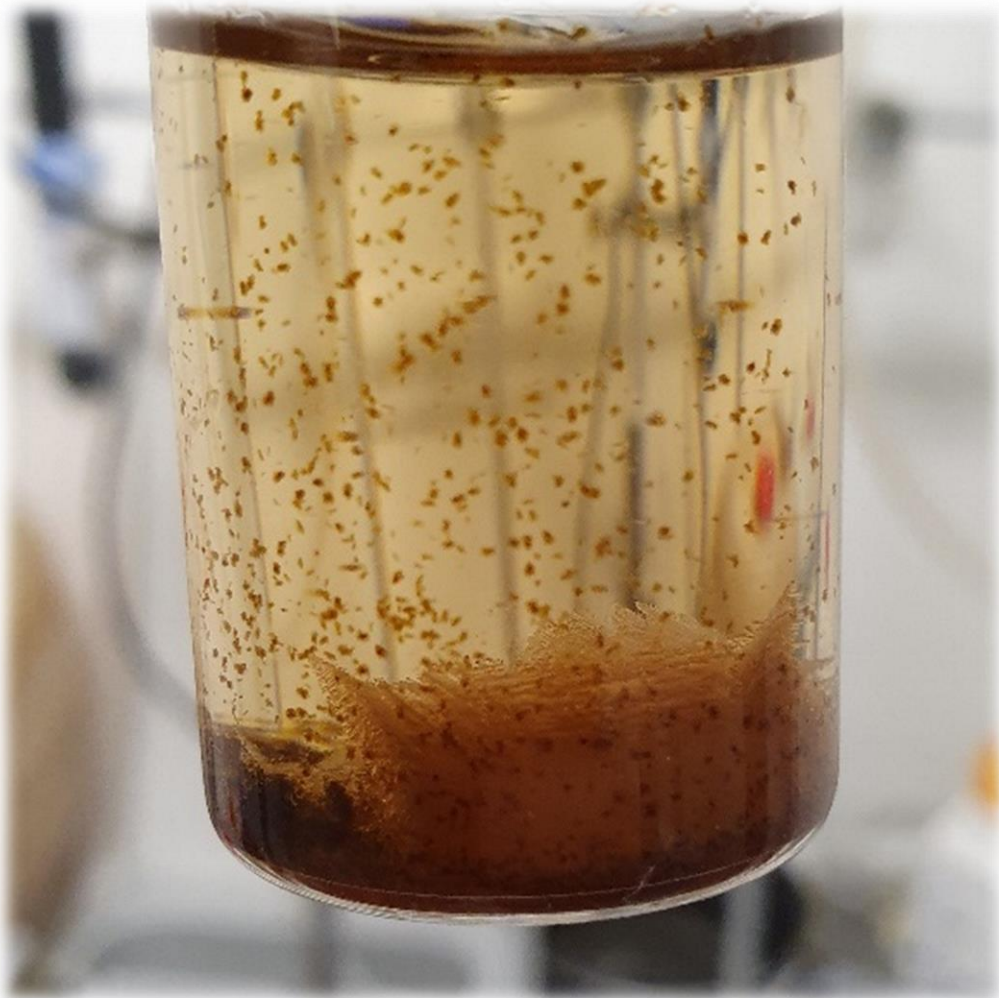
Cubic Crystals

$$\alpha_V = \frac{C_V \langle \gamma \rangle}{V B} \quad \leftarrow \quad \gamma_\lambda = - \frac{V}{\omega_\lambda} \frac{d\omega_\lambda}{dV} \quad \text{Anharmonic Effect!}$$

Non-Cubic Crystals

$$\alpha_{ij} = \frac{C_V}{V} S_{ijkl} \langle \gamma_{kl} \rangle \quad \leftarrow \quad \gamma_{kl}^\lambda = - \frac{1}{\omega_\lambda} \frac{d\omega_\lambda}{d\varepsilon_{kl}} \quad \text{Anharmonic Effect!}$$

Real-Life MOF-74(Zn)



Large mass mismatch between metals and linkers leads to complex phonon properties

Heavier metals shift phonons to lower frequencies

More complex linkers introduce more low-lying linker deformation modes

Inter-system trends of Γ -phonon frequencies can be observed based on classical arguments

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Maxima of Young's Modulus in MOF-74(Zn) are found along the (inclined) zigzag directions

(The first layer of) water increases mostly Young's Modulus along the pore

The trend of ion exchange can be estimated with a simple spring model

Individual deviations from the trends must be analysed in detail exploiting atomic deformations

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Thermal expansion in MOF-74(Zn) is extremely low (XRD and Grüneisen Theory)

Most relevant contributions to the thermal expansion can be found for phonons < 3 THz

The acoustic phonons (especially in Γ_F and Γ_P) show the most pronounced anharmonicities

In preparation